

# **IIT-JEE (ADVANCED)-2023 (PAPER-1)**

### **QUESTIONS WITH SOLUTION FOR IIT-JEE (Advanced) 2023 HELD ON 04 JUNE 2023**

### **PART-I: PHYSICS**

**SECTION-1: (Maximum Marks: 12)** 

- This section contains **THREE** (03) questions.
- Each question has **FOUR** options (A), (B), (C) and (D). **ONE OR MORE THAN ONE** of these four option(s) is(are) correct answer(s).
- For each question, choose the option(s) corresponding to (all) the correct answer(s).
- Answer to each question will be evaluated according to the following marking scheme:

Full Marks : +4 ONLY if (all) the correct option(s) is(are) chosen;

Partial Marks : +3 If all the four options are correct but **ONLY** three options are chosen;

Partial Marks : +2 If three or more options are correct but **ONLY** two options are chosen,

both of which are correct;

Partial Marks : +1 If two or more options are correct but **ONLY** one option is chosen and it

is a correct option;

Zero Marks : 0 If none of the options is chosen (i.e. the question is unanswered);

Negative Marks : -2 In all other cases.

• For example, in a question, if (A), (B) and (D) are the ONLY three options corresponding to correct answers, then

choosing ONLY (A), (B) and (D) will get +4 marks;

choosing ONLY (A) and (B) will get +2 marks;

choosing ONLY (A) and (D) will get +2 marks;

choosing ONLY (B) and (D) will get +2 marks;

choosing ONLY (A) will get +1 marks;

choosing ONLY (B) will get +1 marks;

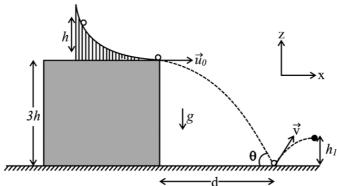
choosing ONLY (D) will get +1 marks;

choosing no option (i.e. the question is unanswered) will get 0 marks; and

choosing any other combination of options will get –2 marks.

### IIT-JEE ADVANCED 2023 Paper-1

1. A slide with a frictionless curved surface, which becomes horizontal at its lower end,, is fixed on the terrace of a building of height 3h from the ground, as shown in the figure. A spherical ball of mass m is released on the slide from rest at a height h from the top of the terrace. The ball leaves the slide with a velocity  $\vec{u}_0 = u_0 \hat{x}$  and falls on the ground at a distance d from the building making an angle  $\theta$  with the horizontal. It bounces off with a velocity  $\vec{v}$  and reaches a maximum height  $h_1$ . The acceleration due to gravity is g and the coefficient of restitution of the ground is  $1/\sqrt{3}$ . Which of the following statement(s) is(are) correct?

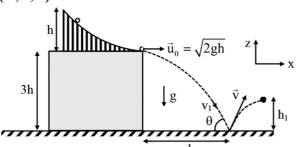


(A) 
$$\vec{u}_0 = \sqrt{2gh}\hat{x}$$

(B) 
$$\vec{v} = \sqrt{2gh} (\hat{x} - \hat{z})$$
 (C)  $\theta = 60^{\circ}$ 

(D) 
$$d/h_1 = 2\sqrt{3}$$

Ans. (A,C,D)



$$\vec{v}_1 = \sqrt{2gh} \hat{i} - \sqrt{2g3h} \hat{k}$$

$$\vec{v} = \sqrt{2gh} \hat{i} + \sqrt{2g3h} \times \frac{1}{\sqrt{3}} \hat{k}$$

$$= \sqrt{2gh} \; \hat{i} + \sqrt{2gh} \; \hat{k}$$

$$\tan \theta = \frac{\sqrt{2g3h}}{\sqrt{2gh}} = \sqrt{3}$$
  $\theta = 60^{\circ}$ 

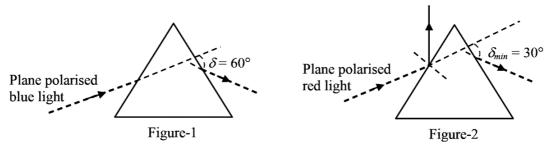
$$h_1 = \frac{v_{1y}^2}{2g} = \frac{2gh}{2g} = h$$

$$d = v_x t = \sqrt{2gh} \times \sqrt{\frac{2 \times 3h}{g}}$$

$$=\sqrt{2gh}\sqrt{\frac{6h}{g}}=2\sqrt{3}h$$

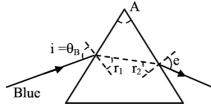
$$=\frac{d}{h_1}=2\sqrt{3}$$

2. A plane polarized blue light ray is incident on a prism such that there is no reflection from the surface of the prism. The angle of deviation of the emergent ray is  $\delta = 60^{\circ}$  (see Figure-1). The angle of minimum deviation for red light from the same prism is  $\delta_{min} = 30^{\circ}$  (see Figure-2). The refractive index of the prism material for blue light is  $\sqrt{3}$ . Which of the following statement(s) is(are) correct?



- (A) The blue light is polarized in the plane of incidence.
- (B) The angle of the prism is 45°.
- (C) The refractive index of the material of the prism for red light is  $\sqrt{2}$ .
- (D) The angle of refraction for blue light in air at the exit plane of the prism is 60°.

### Ans. (A,C,D)



Sol.

$$\tan \theta_{\rm B} = \mu_{\rm B} = \sqrt{3}$$

$$i = \theta_{\rm B} = 60^{\circ}$$

$$1 \sin 60^{\circ} = \sqrt{3} \sin r_{\rm 1}$$

$$r_{\rm 1} = 30^{\circ}$$

$$r_{\rm 1} + r_{\rm 2} = A$$

$$\delta = (i + e) - A$$

$$60^{\circ} = 60^{\circ} + e - A$$

$$e = A$$

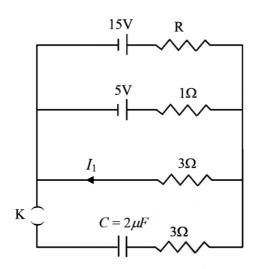
$$\sqrt{3} \sin r_{\rm 2} = 1 \sin e$$

$$\sqrt{3} \sin(A - 30) = \sin A$$
Solving
$$A = 60^{\circ}$$

$$\therefore e = 60^{\circ}$$
For red light
$$\mu = \frac{\sin \left(\frac{A + \delta_{\rm min}}{2}\right)}{\sin A} = \sqrt{2}$$

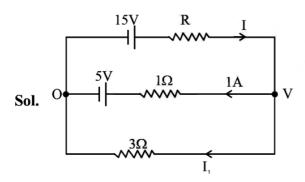
3. In a circuit shown in the figure, the capacitor C is initially uncharged and the key K is open. In this condition, a current of 1 A flows through the 1  $\Omega$  resistor. The key is closed at time  $t = t_0$ . Which of the following statement(s) is(are) correct?

[Given: 
$$e^{-1} = 0.36$$
]



- (A) The value of the resistance R is  $3\Omega$ .
- (B) For  $t < t_0$ , the value of current  $I_1$  is 2A.
- (C) At  $t = t_0 + 7.2 \mu s$ , the current in the capacitor is 0.6 A.
- (D) For  $t \to \infty$ , the charge on the capacitor is 12  $\mu$ C.

#### **Ans.** (**A,B,C,D**)



By writing voltage drop across  $1\Omega$ 

$$\Rightarrow$$
 0 + 5 + 1 × 1 = V

$$V = 6$$

⇒ Similarly across R

$$0+15-I\times R=6$$

$$IR = 9$$

 $\Rightarrow$  across  $3\Omega$ 

$$6 - 3 I_1 = 0$$

$$I_1 = 2A$$

Hence option (B) is correct

$$\Rightarrow$$
 I = 1 + 2

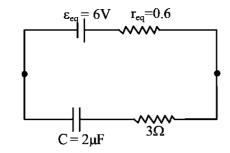
(by KCL)

$$I = 3$$

$$IR = 9$$

$$R = 3\Omega$$

Option (A) is correct



$$\varepsilon = \frac{\frac{15}{3} + \frac{5}{1} + \frac{0}{3}}{\frac{1}{3} + \frac{1}{1} + \frac{1}{3}} = 10 \times \frac{3}{5} = 6V$$

$$q_{max} = 2 \times 6 = 12 \mu C$$

$$i = \frac{6}{3.6} e^{-\frac{t}{\tau}}$$

$$=\frac{5}{3}e - \frac{7.2}{7.2} = \frac{5}{3}e^{-1} \approx 0.6A$$

#### **SECTION-2: (Maximum Marks: 12)**

• This section contains **FOUR (04)** questions.

• Each question has **FOUR** options (A), (B), (C) and (D). **ONLY ONE** of these four options is the correct answer.

• For each question, choose the option corresponding to the correct answer.

• Answer to each question will be evaluated according to the following marking scheme:

Full Marks : +3 If **ONLY** the correct option is chosen;

Zero Marks : 0 If none of the options is chosen (i.e. the question is unanswered);

Negative Marks : -1 In all other cases.

4. A bar of mas M = 1.00 kg and length L = 0.20 m is lying on a horizontal frictionless surface. One end of the bar is pivoted at a point about which it is free to rotate. A small mass m = 0.10 kg is moving on the same horizontal surface with 5.00 m s<sup>-1</sup> speed on a path perpendicular to the bar. It hits the bar at a distance L/2 from the pivoted end and returns back on the same path with speed v. After this elastic collision, the bar rotates with an angular velocity  $\omega$ . Which of the following statement is correct?

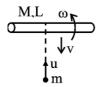
(A) 
$$\omega = 6.98 \text{ rad s}^{-1} \text{ and v} = 4.30 \text{ m s}^{-1}$$

(B) 
$$\omega = 3.75 \text{ rad s}^{-1} \text{ and v} = 4.30 \text{ m s}^{-1}$$

(C) 
$$\omega = 3.75 \text{ rad s}^{-1} \text{ and v} = 10.0 \text{ m s}^{-1}$$

(D) 
$$\omega = 6.80 \text{ rad s}^{-1} \text{ and v} = 4.10 \text{ m s}^{-1}$$

Ans. (A)



Sol.

Applying angular momentum conservation about hinge

$$mv\frac{L}{2} + 0 = -mv\frac{L}{2} + \frac{ML^2}{3}\omega$$
 ....(i)

Also from eq. of restitution

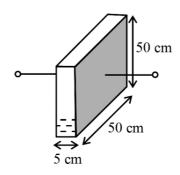
$$e = 1 = \frac{\omega \frac{L}{2} + V}{u} \Rightarrow u = \omega \frac{L}{2} + V \dots (ii)$$

Solving (i) & (ii)

 $\omega \approx 6.98 \text{ rad/sec } \& v = 4.30 \text{ m/s}$ 

Hence option (A)

5. A container has a base of 50 cm  $\times$  5 cm and height 50 cm, as shown in the figure. It has two parallel electrically conducting walls each of area 50 cm  $\times$  50 cm. The remaining walls of the container are thin and non-conducting. The container is being filled with a liquid of dielectric constant 3 at a uniform rate of 250 cm<sup>3</sup> s<sup>-1</sup>. What is the value of the capacitance of the container after 10 seconds? [Given: Permittivity of free space  $\epsilon_0 = 9 \times 10^{-12} \text{ C}^2 \text{ N}^{-1} \text{m}^{-2}$ , the effects of the non-conducting walls on the capacitance are negligible]



(A) 27 pF

(B) 63 pF

(C) 81 pF

(D) 135 pF

Ans. (B)

**Sol.** In t = 10 sec volume of liquid is

$$V = 2500 cc$$

$$h = \frac{2500}{50 \times 5} = 10 \text{cm}$$

$$C_d = \frac{A_d \epsilon_0 k}{d}$$

$$=\frac{50\times10^{-2}\times10\times10^{-2}\epsilon_{0}\times3}{5\times10^{-2}}=3\epsilon_{0}$$

$$C_{_{a}} = \frac{A_{_{a}}\epsilon_{_{0}}}{d} = \frac{50 \times 10^{^{-2}} \times 40 \times 10^{^{-2}}\epsilon_{_{0}}}{5 \times 10^{^{-2}}} = 4\epsilon_{_{0}}$$

$$C = C_a + C_d = 7\epsilon_0$$

$$= 7 \times 9 \times 10^{-12} = 63 \text{ Pf}$$

6. One mole of an ideal gas expands adiabatically from an initial state  $(T_A, V_0)$  to final state  $(T_f, 5V_0)$ . Another mole of the same gas expands isothermally from a different initial state  $(T_B, V_0)$  to the same final state  $(T_f, 5V_0)$ . The ratio of the specific heats at constant pressure and constant volume of this ideal gas is γ. What is the ratio  $T_A/T_B$ ?

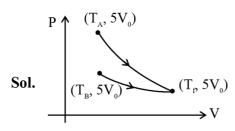
(A)  $5^{\gamma-1}$ 

(B)  $5^{1-\gamma}$ 

(C)  $5^{\gamma}$ 

(D)  $5^{1+\gamma}$ 

Ans. (A)



$$T_A V_0^{\gamma - 1} = T_f (5V_0)^{\gamma - 1}$$

$$\frac{T_A}{T_f} = 5^{\gamma - 1} = \frac{T_A}{T_B}$$

7. Two satellites P and Q are moving in different circular orbits around the Earth (radius R). The heights of P and Q from the Earth surface are  $h_P$  and  $h_Q$ , respectively, where  $h_p = R/3$ . The accelerations of P and Q due to Earth's gravity are  $g_P$  and  $g_Q$ , respectively. If  $g_P/g_Q = 36/25$ , what is the value of  $h_Q$ ?

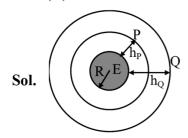
(A) 3R/5

(B) R/6

(C) 6R/5

(D) 5R / 6

Ans. (A)



$$\frac{g_{_P}}{g_{_Q}} = \frac{\frac{GM}{r_{_P}^2}}{\frac{GM}{r_{_Q}^2}} = \left(\frac{r_{_Q}}{r_{_P}}\right)^2$$

$$\frac{36}{25} = \left(\frac{r_Q}{r_P}\right)^2$$

$$\frac{\mathrm{r_Q}}{\mathrm{r_p}} = \frac{6}{5}$$

$$r_{Q} = \frac{6}{5} r_{P}$$

$$R + h_Q = \frac{6}{5} \left( R + \frac{R}{3} \right)$$

$$h_Q = \frac{24}{15}R - R = \frac{9}{15}R = \frac{3}{5}R$$

# IIT-JEE ADVANCED 2023. Paper-1

### **SECTION-3: (Maximum Marks: 24)**

- This section contains **SIX** (**06**) questions.
- The answer to each question is a **NON-NEGATIVE INTEGER**.
- For each question, enter the correct integer corresponding to the answer using the mouse and the on-screen virtual numeric keypad in the place designated to enter the answer.
- Answer to each question will be evaluated according to the following marking scheme:

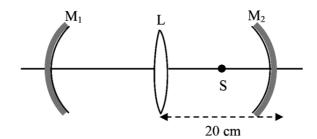
Full Marks : +4 If **ONLY** the correct integer is entered;

Zero Marks : 0 In all other cases.

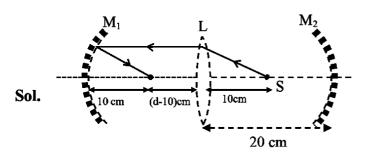
8. A Hydrogen-like atom has atomic number Z. Photons emitted in the electronic transitions from level n=4 to level n=3 in these atoms are used to perform photoelectric effect experiment on a target metal. The maximum kinetic energy of the photoelectrons generated is 1.95 eV. If the photoelectric threshold wavelength for the target metal is 310 nm, the value of Z is \_\_\_\_\_. [Given: hc = 1240 eV-nm and Rhc = 13.6 eV, where R is the Rydberg constant, h is the Planck's constant and c is the speed of light in vacuum]

Ans. (3)

Sol. 
$$n = 4$$
  $n = 3$   
 $-1.51Z^2eV$   $-0.85 Z^2 eV$   
 $E = E_4 - E_3 = 0.66 Z^2 eV$   
 $K_{max} = E - W$   
 $0.66 Z^2 1.95 + 4 = 5.95$   
 $W = 0.66Z^2 - 1.95 = \frac{hc}{\lambda} = \frac{1240}{310}$   
 $\therefore Z = 3$ 



Ans. (80 or 150 or 220)



### Two cases are possible if Ist refraction on lens:

Since object is at focus  $\Rightarrow$  light will become parallel.

Ist reflection at M1:-

Light is parallel  $\Rightarrow$  Image will be at focus.

II<sup>nd</sup> refraction from L:-

$$u = -(d-10)$$

$$f = 10 \text{ cm}$$

$$\frac{1}{x} - \frac{1}{x} = \frac{1}{4}$$

$$\frac{1}{v} + \frac{1}{d-10} = \frac{1}{10}$$

$$\frac{1}{v} = \frac{1}{10} - \frac{1}{(d-10)}$$
 ... (i)

This v will be object for M<sub>2</sub>, and image should be at 10 cm

$$\frac{1}{\mu} + \frac{1}{\mathbf{v}_1} = \frac{1}{\mathbf{f}}$$

$$-\frac{1}{(20-v)}-\frac{1}{10}=-\frac{1}{12}$$

$$\frac{1}{12} - \frac{1}{10} = \frac{1}{20 - v}$$

$$-\frac{2}{120} = \frac{1}{20 - v}$$

$$20 - v = -60$$

$$v = 80 \text{ cm}$$

From equation (i)

$$\frac{1}{80} = \frac{1}{10} - \frac{1}{d-10}$$

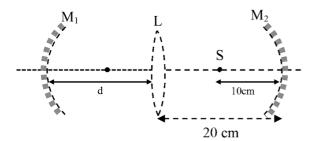
$$\frac{1}{d-10} = \frac{1}{10} - \frac{1}{80}$$

$$\frac{1}{d-10} = \frac{80-10}{800} = \frac{70}{800}$$

$$d-10 = \frac{80}{7} \Rightarrow d = 10 + \frac{80}{7} = \frac{150}{7}$$

$$n = 150$$

# Case-2: If $1^{st}$ reflection on mirror $m_2$



For m<sub>2</sub>

$$\frac{1}{V_1} + \frac{1}{-10} = \frac{1}{-12}$$

$$V_1 = 60 \text{ cm}$$

Then refraction on lens L

$$u_2 = -80 \text{ cm}$$

$$\frac{1}{V_2} - \frac{1}{-60} = \frac{1}{10}$$

$$V_2 = \frac{80}{7}$$

Then reflection on m<sub>2</sub>

Either V<sub>2</sub> is at centre (normal incidence)

$$d-\frac{80}{7}=20$$

$$d = \frac{220}{7}$$

$$\frac{n}{7} = \frac{220}{7},$$

$$n = 220$$

V<sub>2</sub> is at pole of m<sub>2</sub>

$$d-\frac{80}{7}=0$$

$$d = \frac{80}{7}$$

$$\frac{n}{7} = \frac{80}{7}$$

$$n = 80$$

In an experiment for determination of the focal length of a thin convex lens, the distance of the 10. object from the lens is  $10 \pm 0.1$  cm and the distance of its real image from the lens is  $20 \pm 0.2$  cm. The error in the determination of focal length of the lens is n %. The value of n is \_\_\_\_\_.

Ans. (1)

**Sol.** 
$$u = 10 \pm 0.1$$
 cm,  $v = 20 \pm 0.2$  cm

$$\frac{1}{v} - \frac{1}{u} = \frac{1}{f} \Rightarrow \frac{1}{v^2} dv + \frac{1}{u^2} du = -\frac{1}{f^2} df$$

$$\frac{1}{20} + \frac{1}{10} = \frac{1}{f} \Rightarrow \frac{1}{f} = \frac{3}{20} \Rightarrow f = \frac{20}{3} \text{ cm}$$

$$\Rightarrow \frac{1}{(20)^2}(0.2) + \frac{1}{(10)^2}(0.1) = \frac{9}{400} df$$

$$df = \frac{1}{9} \left( \frac{400}{400} \times 0.2 + \frac{400}{100} \times 0.1 \right)$$

$$df = \frac{1}{9} (0.2 + 0.4) \Rightarrow df = \frac{0.6}{9}$$

$$\frac{df}{f} = \frac{0.6}{9} \times \frac{3}{20} = \frac{1}{100}$$

% error = 1 %

**Sol.** 
$$\frac{1}{V} - \frac{1}{U} = \frac{1}{f}$$
;  $+ \frac{1}{20} + \frac{1}{10} = \frac{1}{f}$ 

$$+\frac{1}{20}+\frac{1}{10}=\frac{1}{f}$$

$$-\frac{1}{V^2}dv + \frac{dU}{u^2} = -\frac{df}{f^2} \qquad \frac{1+2}{20} = \frac{1}{f}; \ f = \frac{20}{3}$$

$$\frac{1+2}{20} = \frac{1}{f}$$
;  $f = \frac{20}{3}$ 

$$\frac{0.1}{100} + \frac{0.2}{400} = \frac{6\%}{6}$$

$$\frac{0.4+0.2}{400} = \frac{\Delta f}{f\left(\frac{20}{3}\right)}$$

$$\frac{0.6 \times 20}{400 \times 3} = \frac{\Delta f}{f}$$

$$\frac{1}{100} = \frac{\Delta f}{f}$$

% change in f is 1%

11. A closed container contains a homogeneous mixture of two moles of an ideal monatomic gas  $(\gamma = 5/3)$  and one mole of an ideal diatomic gas  $(\gamma = 7/5)$ . Here,  $\gamma$  is the ratio of the specific heats at constant pressure and constant volume of an ideal gas. The gas mixture does a work of 66 Joule when heated at constant pressure. The change in its internal energy is Joule.

Ans. (121)

Sol. At constant pressure

$$W = nR\Delta T = 66$$

$$\Delta U = n(C_V)_{mix} \Delta T$$

$$(C_{V})_{mix} = \frac{n_{1}C_{V_{1}} + n_{2}C_{V_{2}}}{n_{1} + n_{2}}$$

$$(C_{v})_{mix} = \frac{2 \times \frac{3}{2}R + 1 \times \frac{5}{2}R}{3}$$

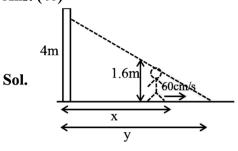
$$\left(C_{V}\right)_{mix} = \frac{11}{6}R$$

$$\Delta U = \frac{11}{6} (nR\Delta T)$$

$$\Delta U = \frac{11}{6} \times 66 = 121J$$

12. A person of height 1.6 m is walking away from a lamp post of height 4 m along a straight path on the flat ground. The lamp post and the person are always perpendicular to the ground. If the speed of the person is 60 cm s<sup>-1</sup>, the speed of the tip of the person's shadow on the ground with respect to the person is \_\_\_\_\_ cm s<sup>-1</sup>.

Ans. (40)



$$\frac{4}{y} = \frac{1.6}{y - x}$$

$$4y - 4x = 1.6y$$

$$2.4 y = 4x$$

$$X = 0.6y$$

$$\frac{dx}{dt} = 0.6 \times \frac{dy}{dt}$$

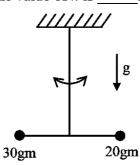
$$60 = 0.6 \times \frac{\mathrm{dy}}{\mathrm{dt}}$$

$$\therefore \frac{\mathrm{dy}}{\mathrm{dt}} = 100 \,\mathrm{cm} \,/\,\mathrm{s}$$

Speed of tip of person's

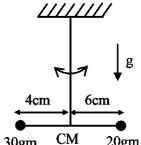
Shadow w.r.t person = 100 - 60 = 40 cm/s

13. Two point-like objects of masses 20 gm and 30 gm are fixed at the two ends of a rigid massless rod of length 10 cm. This system is suspended vertically from a rigid ceiling using a thin wire attached to its center of mass, as shown in the figure. The resulting torsional pendulum undergoes small oscillations. The torsional constant of the wire is  $1.2 \times 10^{-8}$  N m rad<sup>-1</sup>. The angular frequency of the oscillations in  $n \times 10^{-3}$  rad s<sup>-1</sup>. The value of n is



Ans. (10)

Sol.



$$T = \frac{2\pi}{\Omega} = 2\pi \sqrt{\frac{I}{C}}$$

$$\Rightarrow \omega = \sqrt{\frac{C}{I}}$$

Where I = moment of inertia

$$I = (30) (4)^2 + (20) (6)^2$$

$$= 1200 \text{ gm-cm}^2$$

$$= 1.2 \times 10^{-4} \text{ kg-m}^2$$

$$\Rightarrow \omega = \sqrt{\frac{1.2 \times 10^{-8}}{1.2 \times 10^{-4}}}$$

$$\Rightarrow \omega = \sqrt{10^{-4}}$$

$$\omega = (10^{-2})$$

$$n \times 10^{-3} = 10^{-2} \Rightarrow n = 10$$

#### **SECTION-4: (Maximum Marks: 12)**

- This section contains FOUR (04) Matching List Sets.
- Each set has **ONE** Multiple Choice Question.
- Each set has TWO lists: List-I and List-II.
- List-I has Four entries (P), (Q), (R) and (S) and List-II has Five entries (1), (2), (3), (4) and (5).
- FOUR options are given in each Multiple Choice Question based on List-I and List-II and ONLY ONE of these four options satisfies the condition asked in the Multiple Choice Question.
- Answer to each question will be evaluated <u>according to the following marking scheme</u>:

Full Marks : +3 ONLY if the option corresponding to the correct combination is chosen;

Zero Marks : 0 If none of the options is chosen (i.e. the question is unanswered);

Negative Marks : -1 In all other cases.

**14.** List-I shows different radioactive decay processes and List-II provides possible emitted particles. Match each entry in List-I with an appropriate entry from List-II, and choose the correct option.

#### List-I

- (P)  $^{238}_{92}U \rightarrow ^{234}_{91}Pa$
- (Q)  $^{214}_{82}Pb \rightarrow ^{210}_{82}Pb$
- (R)  $^{210}_{81}Tl \rightarrow ^{206}_{82}Pb$
- (S)  $^{228}_{91}Pa \rightarrow ^{224}_{88}Ra$

#### List-II

- (1) one  $\alpha$  particle and one  $\beta^+$  particle
- (2) three  $\beta^-$  particles and one  $\alpha$  particle
- (3) two  $\beta^-$  particles and one  $\alpha$  particle
- (4) one  $\alpha$  particle and one  $\beta^-$  particle
- (5) one  $\alpha$  particle and two  $\beta^+$  particles

(A) 
$$P \rightarrow 4$$
,  $Q \rightarrow 3$ ,  $R \rightarrow 2$ ,  $S \rightarrow 1$ 

(B) 
$$P \rightarrow 4$$
,  $Q \rightarrow 1$ ,  $R \rightarrow 2$ ,  $S \rightarrow 5$ 

(C) 
$$P \rightarrow 5$$
,  $Q \rightarrow 3$ ,  $R \rightarrow 1$ ,  $S \rightarrow 4$ 

(D) 
$$P \rightarrow 5$$
,  $Q \rightarrow 1$ ,  $R \rightarrow 3$ ,  $S \rightarrow 2$ 

Ans. (A)

**Sol.** 
$$_{Z_1}Z^{A_1} \rightarrow_{Z_2} Y^{A_2} + N_{12}He^4 + N_{21}e^0 + N_{3-1}e^0$$

Conservation of charge

$$Z_1 = Z_2 + 2 N_1 + N_2 - N_3$$
 ... (i)

Conservation of nucleons.

$$A_1 = A_2 + 4N_1$$

$$N_1 = \frac{A_1 - A_2}{4}$$
 ... (ii)

From (i) and (ii)

$$N_2 - N_3 = Z_1 - Z_2 - \left(\frac{A_1 - A_2}{2}\right)$$

$$(P)_{92}U^{238} \rightarrow_{91} Pa^{234}$$

$$N_1 = \frac{238 - 234}{4} = 1 \rightarrow 1\alpha$$

$$N_2 - N_3 = (92 - 91) - (\frac{4}{2}) = -1 \rightarrow 1\beta^{-1}$$

(Q) 
$$_{82}Pb^{214} \rightarrow_{82} Pb^{210}$$
  
 $N_1 = \frac{214 - 210}{4} = 1 \rightarrow 1\alpha$ 

$$N_2 - N_3 = (82 - 82) - \left(\frac{4}{2}\right) = -2 \rightarrow 2\beta^{-1}$$

$$(R)_{81}T\ell^{210} \rightarrow_{82} Pb^{206}$$

$$N_1 = \frac{210 - 206}{4} = 1 \rightarrow 1\alpha$$

$$N_2 - N_3 = (81 - 83) - \frac{4}{2} = -3 \rightarrow 3\beta^2$$

(S) 
$$_{91}$$
Pa $^{228} \rightarrow_{88} Ra^{224}$ 

$$N_1 = \frac{228 - 224}{4} = 1\alpha$$

$$N_2 - N_3 = (91 - 88) - \frac{4}{2} = 1\beta^+$$

**15.** Match the temperature of a black body given in List-I with an appropriate statement in List-II, and choose the correct option.

[Given: Wien's constant as  $2.9 \times 10^{-3}$  m-K and  $\frac{hc}{e} = 1.24 \times 10^{-6}$  V-m]

List-I

#### List-II

(P) 2000 K

(1) The radiation at peak wavelength can lead to emission of photoelectrons from a metal of work function 4 eV

(Q) 3000 K

(2) The radiation at peak wavelength is visible to human eye.

(R) 5000 K

- (3) The radiation at peak emission wavelength will result in the widest central maximum of a single slit diffraction.
- (S) 10000 K
- (4) The power emitted per unit area is 1/16 of that emitted by a blackbody at temperature 6000 K.
- (5) The radiation at peak emission wavelength can be used to image human bones.

(A) 
$$P \rightarrow 3$$
,  $Q \rightarrow 5$ ,  $R \rightarrow 2$ ,  $S \rightarrow 3$ 

(B) 
$$P \rightarrow 3$$
,  $Q \rightarrow 2$ ,  $R \rightarrow 4$ ,  $S \rightarrow 1$ 

(C) 
$$P \rightarrow 3$$
,  $Q \rightarrow 4$ ,  $R \rightarrow 2$ ,  $S \rightarrow 1$ 

(D) 
$$P \rightarrow 1$$
,  $Q \rightarrow 2$ ,  $R \rightarrow 5$ ,  $S \rightarrow 3$ 

Ans. (C)

# IIT-JEE ADVANCED 2023. Paper-1

- Sol.  $\Rightarrow$  For option (P) temperature is minimum hence  $\lambda m$  will be maximum  $\beta = \frac{\lambda D}{d} \Rightarrow \beta$  will also be maximum
  - $\Rightarrow$  For option (Q) T = 3000

$$\lambda m = \frac{b}{T} = \frac{2.9 \times 10^{-3}}{30000}$$

$$\lambda m = \frac{2.9}{3} \times 10^{-6}$$

$$=0.96\times10^{-6}$$

$$P_{3000} = 6A (3000)^4$$

$$P_{6000} = 6A (6000)^4$$

$$\frac{P_{3000}}{P_{6000}} = \left(\frac{1}{2}\right)^4 = \frac{1}{16}$$

$$P_{3000} = \frac{1}{16} P_{6000}$$

$$Q-4$$

$$\Rightarrow$$
 For (R) T = 5000 K

$$\lambda m = \frac{2.9 \times 10^{-3}}{5 \times 10^{3}} = 0.58 \times 10^{-6}$$

$$= 580 \text{ nm}$$

Visible to human eyes

$$R-2$$

$$\Rightarrow$$
 For (S) T = 10,000  $\rightarrow$  maximum

Hence (3) is wrong as it has minimum ( $\lambda m$ )

16. A series LCR circuit is connected to a 45 sin ( $\omega t$ ) Volt source. The resonant angular frequency of the circuit is  $10^5$  rad s<sup>-1</sup> and current amplitude at resonance is  $I_0$ . When the angular frequency of the source is  $\omega = 8 \times 10^4$  rad s<sup>-1</sup>, the current amplitude in the circuit is 0.05  $I_0$ . If L = 50 mH, match each entry in List-I with an appropriate value from List-II and choose the correct option.

### List-II List-II

- (P)  $I_0 \text{ in mA}$  (1) 44.4
- (Q) The quality factor of the circuit (2) 18
  - The bandwidth of the circuit in rad  $s^{-1}$  (3) 400
- (S) The peak power dissipated at resonance in Watt (4) 2250 (5) 500
- (A)  $P \rightarrow 2$ ,  $Q \rightarrow 3$ ,  $R \rightarrow 5$ ,  $S \rightarrow 1$
- (B)  $P \rightarrow 3$ ,  $Q \rightarrow 1$ ,  $R \rightarrow 4$ ,  $S \rightarrow 2$
- (C)  $P \rightarrow 4$ ,  $Q \rightarrow 5$ ,  $R \rightarrow 3$ ,  $S \rightarrow 1$
- (D)  $P \rightarrow 4$ ,  $Q \rightarrow 2$ ,  $R \rightarrow 1$ ,  $S \rightarrow 5$

Ans. (B)

**Sol.**  $V = 45 \sin \omega t$ ,

$$L = 50 \text{ mH}$$

$$\omega_0 = 10^5 \, \text{rad} \, / \, \text{s} = \frac{1}{\sqrt{LC}} \Rightarrow C = \frac{1}{L\omega_0^2} = \frac{1}{5 \times 10^{-2} \times 10^{10}}$$

$$= 2 \times 10^{-9} \, \text{F}$$

$$I_0 = \frac{45}{R}$$

$$\omega = 8 \times 10^4 \text{ rad/s} = 0.8 \ \omega_0$$

$$I = 0.05I_0 = \frac{I_0}{20} \Longrightarrow Z = 20R$$

$$X_{L} = 8 \times 10^{4} \times 5 \times 10^{-2} \Omega = 4k\Omega$$

$$X_{C} = \frac{1}{8 \times 10^{4} \times 2 \times 10^{-9}} = \frac{1}{16} \times 10^{5} \Omega = \frac{25}{4} k\Omega$$

$$Z^2 = R^2 + (X_C - X_L)^2$$

$$400R^2 = R^2 + \left(\frac{9}{4}k\Omega\right)^2$$

$$R = \frac{\frac{9}{4} k\Omega}{\sqrt{399}} \approx \frac{9}{80} k\Omega = \frac{900}{8} \Omega$$

$$I_0 = \frac{V_0}{R} = \frac{45 \times 8}{900} = \frac{8}{20} A \approx 0.4A = 400 \text{mA}$$

$$Q = \frac{1}{R} \sqrt{\frac{L}{C}} = \frac{8}{900} \sqrt{\frac{5 \times 10^{-2}}{2 \times 10^{-9}}} = \frac{8}{900} \sqrt{25 \times 10^{6}}$$

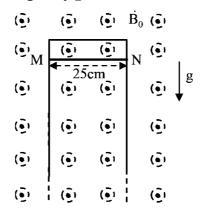
$$Q = \frac{8}{900} \times 5000 = 44.4$$

$$Q = \frac{\omega_0}{\Delta \omega} \Rightarrow \Delta \omega = \frac{\omega_0}{Q} = \frac{10^5}{44.4} = 2250.0$$

$$P_{\text{max}} = I_0^2 R = \frac{45^2}{R^2} \times R = \frac{45^2}{R} = \frac{45^2}{900} \times 8 = 18.4 \text{W}$$

17. A thin conducting rod MN of mass 20 gm, length 25 cm and resistance 10  $\Omega$  is held on frictionless, long, perfectly conducting vertical rails as shown in the figure. There is a uniform magnetic field  $B_0 = 4$  T directed perpendicular to the plane of the rod-rail arrangement. The rod is released from rest at time t = 0 and it moves down along the rails. Assume air drag is negligible. Match each quantity in List-I with an appropriate value from List-II, and choose the correct option.

[Given: The acceleration due to gravity  $g = 10 \text{ ms}^{-2}$  and  $e^{-1} = 0.4$ ]



List-I List-II

- (P) At t = 0.2 s, the magnitude of the induced emf in Volt
  - 0.14 (2)

(1)

- At t = 0.2 s, the magnitude of the magnetic force in Newton (Q)

0.07

- At t = 0.2 s, the power dissipated as heat in Watt (R)
- (3) 1.20
- The magnitude of terminal velocity of the rod in m s<sup>-1</sup> **(S)**
- (4)0.12
- (5)2.00

(A) 
$$P \rightarrow 5$$
,  $Q \rightarrow 2$ ,  $R \rightarrow 3$ ,  $S \rightarrow 1$ 

(B) 
$$P \rightarrow 3$$
,  $Q \rightarrow 1$ ,  $R \rightarrow 4$ ,  $S \rightarrow 5$ 

(C) 
$$P \rightarrow 4$$
,  $Q \rightarrow 3$ ,  $R \rightarrow 1$ ,  $S \rightarrow 2$ 

(D) 
$$P \rightarrow 3$$
,  $Q \rightarrow 4$ ,  $R \rightarrow 2$ ,  $S \rightarrow 5$ 

Ans. (D)

Sol. From force equation

$$mg - Bi\ell = \frac{mdv}{dt}$$

$$mg - \frac{BBi\ell}{R} \times \ell = \frac{mdv}{dt}$$

$$\frac{mgR}{B^2\ell^2} - v = \frac{mR}{B^2\ell^2} \frac{dv}{dt}$$

$$\frac{B^2\ell^2}{mR}\int\limits_{t=0}^t dt = \int\limits_0^v \frac{dv}{\left(\frac{mgR}{B^2\ell^2} - v\right)}$$

Now 
$$\frac{\text{mgR}}{\text{B}^2 \ell^2} = \frac{20 \times 10^{-3} \times 10 \times 10}{16 \times \frac{1}{16}} = 2$$

And 
$$\frac{B^2\ell^2}{mR} = \frac{16 \times \frac{1}{16}}{20 \times 10^{-3} \times 10} = \frac{1}{0.2} = 5$$

$$\therefore 5t = \left[-\ell n \left(2 - v\right)\right]_0^v$$

$$-5t = \ell n \left\lceil \frac{2 - v}{v} \right\rceil$$

$$v = 2 (1 - e^{-5t})$$

$$\therefore v = 2 (1 - e^{-5t})$$
At  $t = 0.2$  sec
$$v = 2 (1 - e^{-5 \times 0.2})$$

$$v = 2 (1 - 0.4)$$

$$v = 1.2 \text{ m/s}$$

(P) Now at t = 0.2 sec

The magnitude of the induced emf =  $E = Bv\ell$ 

$$=4\times1.2\times\frac{1}{4}=1.2\text{Volt}$$

(Q) At t = 0.2 sec, the magnitude of magnetic force =  $BI\ell\sin\theta$ 

$$=B\times\frac{B\ell v}{R}\times\ell\times\sin90^{\circ}$$

$$=\frac{4 \times 4 \times \frac{1}{4} \times 1.3 \times \frac{1}{4}}{10} = 0.12 \text{ Newton}$$

(R) At t = 0.2 sec, the power dissipated as heat

$$P = i^2 R = \frac{v^2}{R} = \frac{1.2 \times 1.2}{10}$$

$$P = 0.144$$
 watt

(S) Magnitude of terminal velocity

At terminal velocity, the net force become zero

$$\therefore$$
 mg = Bi $\ell$ 

$$mg = B {\times} \frac{B\ell v_t}{R} {\times} \ell$$

$$\therefore v_{T} = \frac{mgR}{B^{2}\ell^{2}} = \frac{20 \times 10^{-3} \times 10 \times 10}{16 \times \frac{1}{16}}$$

$$v_T = 2 \text{ m/s}$$

Hence, Answer is (D)



### **PART-II: CHEMISTRY**

### **SECTION-1: (Maximum Marks: 12)**

- This section contains **THREE** (03) questions.
- Each question has **FOUR** options (A), (B), (C) and (D). **ONE OR MORE THAN ONE** of these four option(s) is(are) correct answer(s).
- For each question, choose the option(s) corresponding to (all) the correct answer(s).
- Answer to each question will be evaluated <u>according to the following marking scheme</u>:

Full Marks : +4 ONLY if (all) the correct option(s) is(are) chosen;

Partial Marks : +3 If all the four options are correct but **ONLY** three options are chosen; Partial Marks : +2 If three or more options are correct but **ONLY** two options are chosen,

both of which are correct;

Partial Marks : +1 If two or more options are correct but **ONLY** one option is chosen and it

is a correct option;

Zero Marks : 0 If none of the options is chosen (i.e. the question is unanswered);

Negative Marks : -2 In all other cases.

• For example, in a question, if (A), (B) and (D) are the ONLY three options corresponding to correct answers, then

choosing ONLY (A), (B) and (D) will get +4 marks;

choosing ONLY (A) and (B) will get +2 marks;

choosing ONLY (A) and (D) will get +2 marks;

choosing ONLY (B) and (D) will get +2 marks;

choosing ONLY (A) will get +1 marks;

choosing ONLY (B) will get +1 marks;

choosing ONLY (D) will get +1 marks;

choosing no option (i.e. the question is unanswered) will get 0 marks; and

choosing any other combination of options will get -2 marks.

- 1. The correct statement(s) related to processes involved in the extraction of metals is(are)
  - (A) Roasting of Malachite produces Cuprite.
  - (B) Calcination of Calamine produces Zincite.
  - (C) Copper pyrites is heated with silica in a reverberatory furnace to remove iron.
  - (D) Impure silver is treated with aqueous KCN in the presence of oxygen followed by reduction with zinc metal.

**Ans.** (B,C,D)



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**Sol.** ⇒ Under roasting condition, the malachite will be converted into

$$CuCO_3.Cu(OH)_2 \rightarrow 2CuO + CO_2 + H_2O$$

$$\Rightarrow \operatorname{ZnCO_3} \to \operatorname{ZnO} + \operatorname{CO_2} \uparrow$$
(Calamine) (Zincite)

⇒ Copper pyrites is heated in a reverberatory furnace after mixing with silica. In the furnace, iron oxide 'slag of' as iron silicate and copper is produced in the form of copper matte.

$$FeO + SiO_2 \rightarrow FeSiO_3$$
(Slag)

$$\Rightarrow \text{ Ag + KCN + O}_2 + \text{H}_2\text{O} \longrightarrow [\text{Ag(CN)}_2]^- + \text{KOH}$$

$$\downarrow \text{Zn}$$

$$\text{Ag } \downarrow + [\text{Zn(CN)}_4]^{2^-}$$

2. In the following reactions, P, Q, R, and S are the major products.

$$CH_{3}CH_{2}CH(CH_{3})CH_{2}CN \xrightarrow{(i) PhMgBr, then H_{3}O^{\bigoplus}} \mathbf{P}$$

Ph – H + CH<sub>3</sub>CCl 
$$\xrightarrow{\text{(i) anhyd. AlCl}_3}$$
 Q

$$CH_{3}CH_{2}CC1 \xrightarrow{(i)\frac{1}{2}(PhCH_{2})_{2}Cd} R$$

$$PhCH_{2}CHO \xrightarrow{\begin{subarray}{c} (i) PhMgBr, then $H_{2}O$\\ \hline (ii) CrO_{3}, dil. $H_{2}SO_{4}$\\ \hline (iii) HCN\\ (iv) H_{2}SO_{4}, \Delta \end{subarray}} S$$

The correct statement(s) about P, Q, R, and S is(are)

- (A) Both **P** and **Q** have asymmetric carbon(s).
- (B) Both **Q** and **R** have asymmetric carbon(s).
- (C) Both **P** and **R** have asymmetric carbon(s).
- (D) P has asymmetric carbon(s), S does not have any asymmetric carbon.

Ans. (C,D)

#### Sol. Formation of P

$$CH_3 - CH_2 - CH - CH_2 - CN \xrightarrow{PhMgBr} H_3O^+$$

$$PhMgBr \\ then H_3O^+$$

$$Asymmetric \\ carbon \\ (P)$$

### Formation of Q

### Formation of R

$$C-C1 + \frac{1}{2} (Ph - CH_2)_2 Cd$$

$$Ph$$

$$PhMgBr$$

$$then H_3 O^+$$

$$asymmetric$$

$$carbon$$

$$(R)$$

#### Formation of S

$$Ph - CH_{2} - C - H \xrightarrow{PhMgBr} Ph - CH_{2} - CH - Ph$$

$$OH$$

$$CrO_{3} \text{ with dil. } H_{2}SO_{4}$$

$$Ph - CH_{2} - C - Ph$$

$$HCN$$

$$OH$$

$$OH$$

$$OH$$

$$OH$$

$$COOH$$

$$OH$$

$$COOH$$

(S) No asymmetric carbon

Consider the following reaction scheme and choose the correct option(s) for the major products Q,R and S.

Styrene 
$$\xrightarrow{\text{(i) B}_2\text{H}_6} \xrightarrow{\text{(ii) NaOH, H}_2\text{O}_2, \text{H}_2\text{O}} P \xrightarrow{\text{(ii) CIO}_3, \text{H}_2\text{SO}_4} \xrightarrow{\text{(ii) CI}_2, \text{Red Phosphorus}} Q$$

$$P \xrightarrow{\text{(i) SOCI}_2} \xrightarrow{\text{(ii) NaCN}} R \xrightarrow{\text{conc.H}_2\text{SO}_4} S$$

$$CI \downarrow \text{COOH} \qquad \downarrow \text{NH}_2 \qquad \downarrow \text{NH}_2 \qquad \downarrow \text{COOH} \qquad \downarrow \text{COOH}$$

$$Q \qquad R \qquad S \qquad Q \qquad R \qquad S$$

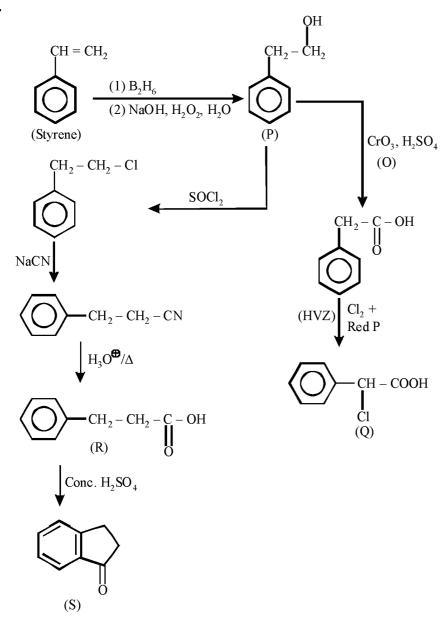
$$Q \qquad R \qquad S \qquad Q \qquad R \qquad S$$

$$Q \qquad R \qquad S \qquad Q \qquad R \qquad S$$

$$Q \qquad R \qquad S \qquad Q \qquad R \qquad S$$

Ans. (B)

Sol.





#### **SECTION-2: (Maximum Marks: 12)**

- This section contains **FOUR (04)** questions.
- Each question has FOUR options (A), (B), (C) and (D). ONLY ONE of these four options is the correct answer.
- For each question, choose the option corresponding to the correct answer.
- Answer to each question will be evaluated according to the following marking scheme:

Full Marks : +3 If **ONLY** the correct option is chosen;

Zero Marks : 0 If none of the options is chosen (i.e. the question is unanswered);

: -1 In all other cases. Negative Marks

4. In the scheme given below, X and Y, respectively, are

Metal halide 
$$\xrightarrow{\text{aq. NaOH}}$$
 White precipitate **(P)** + Filtrate **(Q)**

$$\mathbf{P} \xrightarrow{\text{aq.H}_2SO_4 \atop \text{PbO}_2(\text{excess})} \mathbf{X} \text{ (a coloured species in solution)}$$

$$\mathbf{Q} \xrightarrow{\text{Conc.H}_2SO_4 \atop \text{warm}} \mathbf{Y} \text{ (gives blue-coloration with KI-starch paper)}$$

(A)  $CrO_4^{2-}$  and  $Br_2$ 

(B)  $MnO_4^{2-}$  and  $Cl_2$ 

(C) MnO<sub>4</sub> and Cl<sub>2</sub>

(D) MnSO<sub>4</sub> and HOCl

Ans. (C)

Sol. 
$$MnCl_2 + NaOH \rightarrow Mn(OH)_2 \downarrow + NaCl$$
(P)
(white ppt.) (Filterate)
$$Mn(OH)_2 \xrightarrow{PbO_2 + H^+(H_2SO_4)} MnO_4^- + Pb^{2+}$$
Purple

Plotting  $1/\Lambda_m$  against  $c\Lambda_m$  for aqueous solutions of a monobasic weak acid (HX) resulted in a 5. straight line with y-axis intercept of P and slope of S. The ratio P/S is

 $[\Lambda_{\rm m} = {\rm molar\ conductivity}]$ 

 $\Lambda_{\rm m}^{\circ}$  = limiting molar conductivity

c = molar concentration

 $K_a$  = dissociation constant of HX]

- (A)  $K_a \Lambda_m^{\circ}$

- (B)  $K_a \Lambda_m^{\circ} / 2$  (C)  $2 K_a \Lambda_m^{\circ}$  (D)  $1 / (K_a \Lambda_m^{\circ})$

Ans. (A)

**Sol.** For weak acid, 
$$\alpha = \frac{\Lambda_m}{\Lambda_0}$$

$$K_a = \frac{C\alpha^2}{1-\alpha} \Rightarrow K_a (1-\alpha) = C\alpha^2$$

$$\Rightarrow K_a \left( 1 - \frac{\Lambda_m}{\Lambda_0} \right) = C \left( \frac{\Lambda_m}{\Lambda_0} \right)^2$$

$$\Rightarrow K_a - \frac{\Lambda_m K_a}{\Lambda_0} = \frac{C \Lambda_m^2}{(\Lambda_0)^2}$$

Divide by  $'\Lambda_m'$ 

$$\Rightarrow \frac{K_a}{\Lambda_m} = \frac{C\Lambda_m}{(\Lambda_0)^2} + \frac{K_a}{\Lambda_0}$$

$$\Rightarrow \frac{1}{\Lambda_{\rm m}} = \frac{C\Lambda_{\rm m}}{K_{\rm a}(\Lambda_0)^2} + \frac{1}{\Lambda_0}$$

Plot 
$$\frac{1}{\Lambda_m}$$
 vs  $C\Lambda_m$  has

Slope = 
$$\frac{1}{K_0(\Lambda_0)^2} = S$$

y-intercept = 
$$\frac{1}{\Lambda_0}$$
 = P

Then, 
$$\frac{P}{S} = \frac{\frac{1}{\Lambda_0}}{\frac{1}{K_a(\Lambda_0)^2}} = K_a \Lambda_0$$

- 6. On decreasing the pH from 7 to 2, the solubility of a sparingly soluble salt (MX) of a weak acid (HX) increased from  $10^{-4}$  mol L<sup>-1</sup> to  $10^{-3}$  mol L<sup>-1</sup>. The  $pK_a$  of HX is:
  - (A)3

(B)4

- (C) 5
- (D) 2

Ans. (B)

**Sol.** At pH = 7 
$$\Rightarrow$$
 pure water solubility =  $S_1 = \sqrt{K_{sp}}$   
At pH = 2

$$\Rightarrow MX(s) + aq \xrightarrow{K_{SP}} M^{+}(aq) + X^{-}(aq)$$

$$X^{-}(aq) + H^{+}(aq) \xrightarrow{1/K_a} HX(aq)$$

$$s-x \quad 10^{-2} \qquad x \approx s$$

s-x  $10^{-2}$   $x \approx s$ Approximation :  $s - x \approx 0$  [X<sup>-</sup> is limiting reagent]

$$\Rightarrow s \simeq x$$

$$\Rightarrow s(s-x) = K_{sp} \qquad .....(1)$$

$$\frac{s}{(s-x)(10^{-2})} = \frac{1}{K_a} \qquad .....(2)$$

Multiply (1) × (2) 
$$\Rightarrow \frac{s^2}{10^{-2}} = \frac{K_{sp}}{K_a}$$
  
 $\Rightarrow s = \frac{\sqrt{K_{sp}}}{10\sqrt{K_a}}$ 

Now given : 
$$\frac{s}{s_1} = \frac{10^{-3}}{10^{-4}}$$

$$\Rightarrow \frac{\sqrt{K_{sp}}}{10\sqrt{K_a}} = 10 \qquad \Rightarrow \frac{1}{10\sqrt{K_a}} = 10$$
$$\Rightarrow \sqrt{K_a} = 10^{-2}$$
$$\Rightarrow K_a = 10^{-4}$$
$$\Rightarrow pK_a = 4$$

7. In the given reaction scheme, P is a phenyl alkyl ether, Q is an aromatic compound; R and S are the major products.

$$P \xrightarrow{\text{HI}} Q \xrightarrow{\text{(ii) NaOH} \atop \text{(iii) CO}_2} R \xrightarrow{\text{(i)(CH}_3\text{CO)}_2\text{O}} S$$

The correct statement about S is

- (A) It primarily inhibits noradrenaline degrading enzymes.
- (B) It inhibits the synthesis of prostaglandin.
- (C) It is a narcotic drug.
- (D) It is ortho-acetylbenzoic acid.

Ans. (B)

Sol. P is phenyl alkyl ether

Q is aromatic compound

R and S are the major product

i.e. -RO - HH-I Phenyl alkyl ether (1) NaOH  $O^-Na^+$ COOH COOH (Aspirin) Pain killer Acetyl salicylic acid

Correct ans is (B)

Aspirin inhibits the synthesis of chemicals known as prostaglandin's.

### **SECTION-3: (Maximum Marks: 24)**

(Non-narcotic analgesic)

This section contains SIX (06) questions.

The answer to each question is a **NON-NEGATIVE INTEGER**.

For each question, enter the correct integer corresponding to the answer using the mouse and the on-screen virtual numeric keypad in the place designated to enter the answer.

Answer to each question will be evaluated according to the following marking scheme:

: +4 **ONLY** If the correct integer is entered; Full Marks

Zero Marks In all other cases.

8. The stoichiometric reaction of 516 g of dimethyldichlorosilane with water results in a tetrameric cyclic product **X** in 75% yield. The weight (in g) of **X** obtained is \_\_\_\_. [Use, molar mass  $(g \text{ mol}^{-1})$ : H = 1, C = 12, O = 16, Si = 28, Cl = 35.5]

Ans. (222)

**Sol.** 
$$4(CH_3)_2SiCl_2 + 4H_2O \xrightarrow{75\%} (CH_3)_8Si_4O_4 + 8HCl_{(X)}$$

$$w = 516g$$

$$n 
 (moles) = \frac{516}{129}$$

$$= 4$$

weight = 296 g

$$% yield = 75$$

The weight of X (in gram) = 
$$296 \times \frac{75}{100} = 222 \text{ g}$$

9. A gas has a compressibility factor of 0.5 and a molar volume of 0.4 dm<sup>3</sup> mol<sup>-1</sup> at a temperature of 800 K and pressure **x** atm. If it shows ideal gas behaviour at the same temperature and pressure, the molar volume will be **y** dm<sup>3</sup> mol<sup>-1</sup>. The value of **x/y** is \_\_\_\_.

[Use: Gas constant, 
$$R = 8 \times 10^{-2} L \text{ atm K}^{-1} \text{ mol}^{-1}$$
]

### Ans. (100)

**Sol.** For gas : 
$$Z = 0.5$$
,  $V_m = 0.4$  L/mol

$$T = 800 \text{ K}, P = X \text{ atm}.$$

$$\Rightarrow Z = \frac{PV_m}{RT}$$

$$\Rightarrow \frac{X(0.4)}{0.08 \times 800} = 0.5$$

$$\Rightarrow X = 80$$

For ideal gas,  $PV_m = RT$ 

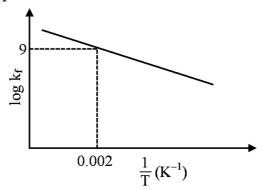
$$\Rightarrow V_{m} = \frac{RT}{P} = \frac{0.08 \times 800}{80} = 0.8 \text{ L mol}^{-1} = y$$

Then, 
$$\frac{x}{y} = \frac{80}{0.8} = 100.$$





10. The plot of log  $k_f$  versus  $\frac{1}{T}$  for a reversible reaction  $A(g) \rightleftharpoons P(g)$  is shown.



Pre-exponential factors for the forward and backward reactions are  $10^{15}$  s<sup>-1</sup> and  $10^{11}$  s<sup>-1</sup>, respectively. If the value of log K for the reaction at 500 K is 6, the value of log k<sub>b</sub> at 250 K is

 $\overline{|K|}$  = equilibrium constant of the reaction

 $k_f$  = rate constant of forward reaction

 $k_b$  = rate constant of backward reaction

Ans. (5)

**Sol.** For reaction  $A(g) \Longrightarrow P(g)$ 

 $\log k_f = \frac{-E_f}{2.303 \, RT} + \log A_f [Arrhenius equation for forward reaction]$ 

From plot when, 
$$\frac{1}{T} = 0.002$$
,  $\log k_f = 9$ 

$$\Rightarrow 9 = \frac{-E_f}{2.303 \,\text{R}} (0.002) + \log (A_f)$$

Given: 
$$A_f = 10^{15} \text{ s}^{-1}$$

$$\Rightarrow$$
 9 =  $\frac{-E_f}{2.303 \,\text{R}}$  (0.002) + 15

$$\Rightarrow \frac{E_f}{2.303R} = \frac{6}{0.002} = 3000$$

Now, 
$$K = \frac{k_f}{k_h} = \frac{A_f}{A_h} e^{-(E_f - E_b)/RT}$$

$$\log K = -\frac{1}{2.303} \frac{(E_f - E_b)}{RT} + \log \left(\frac{10^{15}}{10^{11}}\right)$$

A+ 500 K

$$\Rightarrow$$
 6 =  $\frac{-(E_f - E_b)}{500R (2.303)} + 4$ 

$$\Rightarrow$$
 (1000 R) (2.303) =  $E_b - E_f$ 

$$\Rightarrow$$
 (1000 R) (2.303) = E<sub>b</sub> - 3000 (2.303 R)

$$\Rightarrow E_b = 4000 \text{ R} (2.303)$$

Now  $k_b = A_b e^{-E_b/RT}$ 

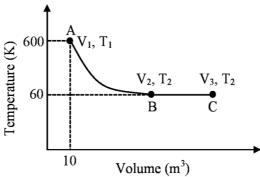
....(1)

$$\Rightarrow \log k_b = \frac{-E_b}{2.303 \,\text{RT}} + \log A_b$$
At 250 K
$$\Rightarrow \log k_b = -\frac{4000}{250} + \log (10^{11})$$
 [From equation (1)]

$$\begin{array}{c}
250 \\
= -16 + 11 = -5
\end{array}$$

 $|\log k_b| = 5$ 

11. One mole of an ideal monoatomic gas undergoes two reversible processes  $(A \rightarrow B \text{ and } B \rightarrow C)$  as shown in the given figure:



 $A \to B$  is an adiabatic process. If the total heat absorbed in the entire process  $(A \to B \text{ and } B \to C)$ is  $RT_2 \ln 10$ , the value of  $2 \log V_3$  is

[Use, molar heat capacity of the gas at constant pressure,  $C_{p,m} = \frac{5}{2}R$ ]

Ans. (7)

**Sol.** For A 
$$\to$$
 B  $600 V_1^{\gamma - 1} = 60 V_2^{\gamma - 1} \ (\gamma = 5/3)$ 

(Reversible adiabatic)

$$\Rightarrow$$
 600  $(V_1)^{2/3} = 60 (V_2)^{2/3}$ 

$$\Rightarrow 10 = \left(\frac{V_2}{V_1}\right)^{2/3}$$

$$\Rightarrow 10 = \left(\frac{V_2}{10}\right)^{2/3}$$

$$\Rightarrow V_2 = 10(10)^{3/2} = 10^{5/2}$$

Now,  $q_{net} = RT_2 \ln 10 = 60 R \ln 10 = q_{AB} + q_{BC}$   $\therefore q_{AB} = 0$ 

$$\therefore$$
  $q_{AB} = 0$ 

$$\Rightarrow$$
 q<sub>BC</sub> = 60 R In 10 = 60 R In  $\frac{V_3}{V_2}$ 

 $[:: B \to C \text{ is reversible isothermal}]$ 

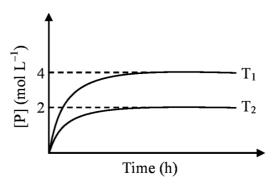
$$\Rightarrow$$
 60 R  $ln$  10 = 60 R  $ln$   $\left(\frac{V_3}{10^{5/2}}\right)$ 

$$\Rightarrow \log 10 = \log V_3 - \frac{5}{2}$$

$$\Rightarrow \log V_3 = \frac{7}{2} \Rightarrow 2 \log V_3 = 7$$



12. In a one-litre flask, 6 moles of A undergoes the reaction A (g)  $\rightleftharpoons$  P (g). The progress of product formation at two temperatures (in Kelvin),  $T_1$  and  $T_2$ , is shown in the figure:



If  $T_1 = 2T_2$  and  $(\Delta G_2^{\Theta} - \Delta G_1^{\Theta}) = RT_2 \ln x$ , then the value of x is \_\_\_\_\_.

[ $\Delta G_1^{\Theta}$  and  $\Delta G_2^{\Theta}$  are standard Gibb's free energy change for the reaction at temperatures  $T_1$  and  $T_2$ , respectively.]

Ans. (8)

**Sol.** At 
$$T_1 K$$
:  $A(g) \Longrightarrow P(g)$ 

$$t = 0$$

$$t = \infty$$
  $6 - x$   $x = 4$  (from plot)

$$\Rightarrow$$
 At T<sub>1</sub> K : K<sub>P<sub>1</sub></sub> =  $\frac{4}{2}$  = 2

At 
$$T_2 K$$
:  $A(g) \Longrightarrow P(g)$ 

$$t = 0$$

$$t = \infty$$
  $6 - y$   $y = 2$  (from plot)

$$\Rightarrow$$
 At T<sub>2</sub> K :  $K_{P_2} = \frac{2}{4} = \frac{1}{2}$ 

Now, 
$$\Delta G_2^0 = -RT_2 \ln K_{P_2} = -RT_2 \ln \frac{1}{2}$$

$$\Rightarrow \Delta G_2^0 = RT_2 \ln 2$$

$$\Delta G_1^0 = -RT_1 \ln K_{P_1} = -RT_1 \ln 2 = -2RT_2 \ln 2$$

Given : 
$$\Delta G_2^o - \Delta G_1^o = RT_2 \ln 2 + 2RT_2 \ln 2 = 3RT_2 \ln 2 = RT_2 \ln x$$

$$\Rightarrow x = 2^3 = 8$$

# IIT-JEE ADVANCED 2023

PAPER-1

13. The total number of  $sp^2$  hybridised carbon atoms in the major product **P** (a non-heterocyclic compound) of the following reaction is \_\_\_\_\_.

$$NC \downarrow_{NC}^{CN} \stackrel{(i) \text{ LiAlH}_4 \text{ (excess), then H}_2O}{\text{(ii) Acetophenone (excess)}} P$$

Ans. (28)

Sol.

Total number of  $sp^2$  hybridised C-atom in P = 28

#### **SECTION-4: (Maximum Marks: 12)**

- This section contains **FOUR (04)** Matching List Sets.
- Each set has **ONE** Multiple Choice Question.
- Each set has **TWO** lists: **List-I** and **List-II**.
- List-I has Four entries (P), (Q), (R) and (S) and List-II has Five entries (1), (2), (3), (4) and (5).
- FOUR options are given in each Multiple Choice Question based on List-I and List-II and ONLY ONE of these four options satisfies the condition asked in the Multiple Choice Question.
- Answer to each question will be evaluated according to the following marking scheme:

Full Marks: +3 ONLY if the option corresponding to the correct combination is chosen;

Zero Marks : 0 If none of the options is chosen (i.e. the question is unanswered);

Negative Marks : -1 In all other cases.

14. Match the reactions (in the given stoichiometry of the reactants) in List-I with one of their products given in List-II and choose the correct option.

#### List-I

(P) 
$$P_2O_3 + 3H_2O \rightarrow$$

(Q) 
$$P_4 + 3NaOH + 3H_2O \rightarrow$$

(R) 
$$PCl_5 + CH_3COOH \rightarrow$$

(S) 
$$H_3PO_2 + 2H_2O + 4AgNO_3 \rightarrow$$

#### List-II

- (1) P(O)(OCH<sub>3</sub>)Cl<sub>2</sub>
- (2)  $H_3PO_3$
- (3) PH<sub>3</sub>
- (4) POCl<sub>3</sub>

$$(5)$$
  $H_3PO_4$ 

(A) 
$$P \rightarrow 2$$
;  $Q \rightarrow 3$ ;  $R \rightarrow 1$ ;  $S \rightarrow 5$ 

(C) 
$$P \rightarrow 5$$
;  $Q \rightarrow 2$ ;  $R \rightarrow 1$ ;  $S \rightarrow 3$ 

(B) 
$$P \rightarrow 3$$
;  $Q \rightarrow 5$ ;  $R \rightarrow 4$ ;  $S \rightarrow 2$ 

(D) P 
$$\rightarrow$$
 2; Q  $\rightarrow$  3; R  $\rightarrow$  4; S  $\rightarrow$  5

Ans. (D)

**Sol.** (P)  $P_2O_3 + 3H_2O \rightarrow 2H_3PO_3$ 

(Q)  $P_4 + 3NaOH + 3H_2O \rightarrow 3NaH_2PO_2 + PH_3$ 

(R)  $PCl_5 + CH_3COOH \rightarrow CH_3COCl + \underline{POCl_3} + HCl$ 

(S)  $H_3PO_2 + 2H_2O + 4AgNO_3 \rightarrow 4Ag + 4HNO_3 + H_3PO_4$ 

15. Match the electronic configurations in List-I with appropriate metal complex ions in List-II and choose the correct option.

[Atomic Number: Fe = 26, Mn = 25, Co = 27]

List-I

- (P)  $t_{2g}^6 e_g^0$
- (Q)  $t_{2g}^3 e_g^2$
- (R)  $e^2t_2^3$
- $(S) t_{2g}^4 e_g^2$
- (A)  $P \rightarrow 1$ ;  $Q \rightarrow 4$ ;  $R \rightarrow 2$ ;  $S \rightarrow 3$
- (C)  $P \rightarrow 3$ ;  $Q \rightarrow 2$ ;  $R \rightarrow 5$ ;  $S \rightarrow 1$

Ans. (D)

Sol. 1.  $[Fe(H_2O)_6]^{+2}$ WFL

configuration 
$$3d^{\frac{6}{1000}}e_g$$

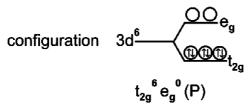
$$t_{2a}^{\frac{4}{1000}}e_g$$

2. [Mn(H<sub>2</sub>O)<sub>6</sub>]<sup>+2</sup>

configuration 
$$3d^{\frac{5}{2}}$$

$$t_{2g}^{\frac{3}{9}}e_{g}^{\frac{2}{9}}(Q)$$

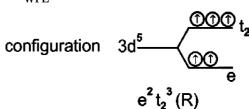
3.  $[Co(NH_3)_6]^{+3}$ 



List-II

- (1)  $[Fe(H_2O)_6]^{2+}$
- (2)  $[Mn(H_2O)_6]^{2+}$
- (3)  $\left[\text{Co(NH_3)}_6\right]^{3+}$
- (4)  $[FeCl_4]^-$
- (5)  $[CoCl_4]^{2-}$
- (B)  $P \rightarrow 1$ ;  $Q \rightarrow 2$ ;  $R \rightarrow 4$ ;  $S \rightarrow 5$
- (D)  $P \rightarrow 3$ ;  $Q \rightarrow 2$ ;  $R \rightarrow 4$ ;  $S \rightarrow 1$

III 4. [FeCl<sub>4</sub>]<sup>Θ</sup> WFL

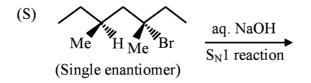


**16.** Match the reactions in List-I with the features of their products in List-II and choose the correct option.

#### List-I

#### List-II

- (P) (-)-1-Bromo-2-ethylpentane  $\frac{\text{aq. NaOH}}{\text{S}_{\text{N}}\text{2 reaction}}$
- (1) Inversion of configuration
- (Q) (-)-2-Bromopentane aq. NaOH (single enantiomer)  $S_N$ 2 reaction
- (2) Retention of configuration
- (R) (-)-3-Bromo-3-methylhexane (single enantiomer)  $\frac{\text{aq. NaOH}}{\text{S}_{\text{N}1} \text{ reaction}}$
- (3) Mixture of enantiomers



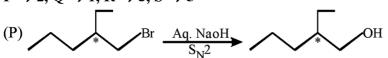
(4) Mixture of structural isomers

(5) Mixture of diastereomers

- (A)  $P \rightarrow 1$ ;  $Q \rightarrow 2$ ;  $R \rightarrow 5$ ;  $S \rightarrow 3$
- (B)  $P \rightarrow 2$ ;  $Q \rightarrow 1$ ;  $R \rightarrow 3$ ;  $S \rightarrow 5$
- (C)  $P \rightarrow 1$ ;  $Q \rightarrow 2$ ;  $R \rightarrow 5$ ;  $S \rightarrow 4$
- (D)  $P \rightarrow 2$ ;  $Q \rightarrow 4$ ;  $R \rightarrow 3$ ;  $S \rightarrow 5$

Ans. (B)

Sol.  $P \rightarrow 2$ ,  $Q \rightarrow 1$ ,  $R \rightarrow 3$ ,  $S \rightarrow 5$ 

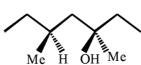


Retention of configuration

(Q) 
$$Aq. NaoH$$
  $S_N^2$  OH

Inversion of configuration

$$(R) \xrightarrow{\text{Aq. NaoH}} \xrightarrow{\text{OH}} \xrightarrow{\text{OH}}$$



Diastereomeric mixture

**(S)** 

17. The major products obtained from the reactions in List-II are the reactants for the named reactions mentioned in List-I. Match List-I with List-II and choose the correct option.

List-I

Etard reaction

List-II

(1) Acetophenone  $Zn-Hg, HCl \rightarrow$ 

(Q) Gattermann reaction

(P)

- (2) Toluene  $(i) \text{ KMnO}_4, \text{KOH, } \Delta \rightarrow (ii) \text{ SOCl}_2$
- (R) Gattermann-Koch reaction
- Benzene  $\xrightarrow{\text{CH}_3\text{Cl}}$  anhyd. AlCl<sub>3</sub>
- (S) Rosenmund reduction
- (4) Aniline  $\frac{\text{NaNO}_2/\text{HCl}}{273-278 \text{ K}}$
- (5) Phenol  $Zn, \Delta \rightarrow$

(A) 
$$P \rightarrow 2$$
;  $Q \rightarrow 4$ ;  $R \rightarrow 1$ ;  $S \rightarrow 3$ 

(B) 
$$P \rightarrow 1$$
;  $Q \rightarrow 3$ ;  $R \rightarrow 5$ ;  $S \rightarrow 2$ 

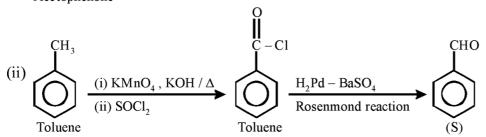
(C) 
$$P \rightarrow 3$$
;  $Q \rightarrow 2$ ;  $R \rightarrow 1$ ;  $S \rightarrow 4$ 

(D) 
$$P \rightarrow 3$$
;  $Q \rightarrow 4$ ;  $R \rightarrow 5$ ;  $S \rightarrow 2$ 

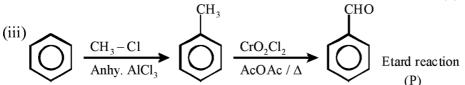
Ans. (D)

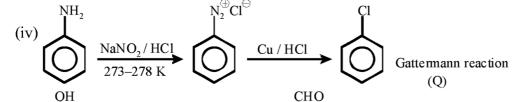
Sol.  $P \rightarrow 3$ ,  $Q \rightarrow 4$ ,  $R \rightarrow 5$ ,  $S \rightarrow 2$ 

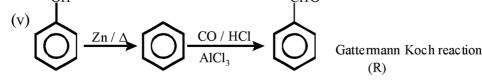
(i) 
$$\prod_{\substack{Ph-C-CH_3\\Acetophenone}} \frac{Zn-Hg/HCl}{Ph-CH_2-CH_2}$$



(3)







### PART-III: MATHEMATICS

#### **SECTION-1: (Maximum Marks: 12)**

• This section contains **THREE** (03) questions.

• Each question has **FOUR** options (A), (B), (C) and (D). **ONE OR MORE THAN ONE** of these four option(s) is(are) correct answer(s).

• For each question, choose the option(s) corresponding to (all) the correct answer(s).

• Answer to each question will be evaluated <u>according to the following marking scheme</u>:

Full Marks : +4 ONLY if (all) the correct option(s) is(are) chosen;

Partial Marks : +3 If all the four options are correct but **ONLY** three options are chosen;

Partial Marks : +2 If three or more options are correct but **ONLY** two options are chosen,

both of which are correct;

Partial Marks : +1 If two or more options are correct but **ONLY** one option is chosen and it

is a correct option;

Zero Marks : 0 If none of the options is chosen (i.e. the question is unanswered);

Negative Marks : -2 In all other cases.

• For example, in a question, if (A), (B) and (D) are the ONLY three options corresponding to correct answers, then

choosing ONLY (A), (B) and (D) will get +4 marks;

choosing ONLY (A) and (B) will get +2 marks;

choosing ONLY (A) and (D) will get +2 marks;

choosing ONLY (B) and (D) will get +2 marks;

choosing ONLY (A) will get +1 marks;

choosing ONLY (B) will get +1 marks;

choosing ONLY (D) will get +1 marks;

choosing no option (i.e. the question is unanswered) will get 0 marks; and

choosing any other combination of options will get -2 marks.

- 1. Let  $S = (0, 1) \cup (1, 2) \cup (3, 4)$  and  $T = \{0, 1, 2, 3\}$ . Then which of the following statements is(are) true?
  - (A) There are infinitely many functions from S to T
  - (B) There are infinitely many strictly increasing functions from S to T
  - (C) The number of continuous functions from S to T is at most 120
  - (D) Every continuous function from S to T is differentiable

Ans. (ACD)

**Sol.**  $S = (0, 1) \cup (1, 2) \cup (3, 4)$ 

$$T = \{0, 1, 2, 3\}$$

Number of functions:

Each element of S have 4 choice

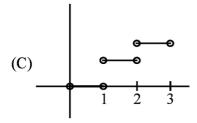
Let n be the number of element in set S.

Number of function  $= 4^n$ 

Here  $n \to \infty$ 

 $\Rightarrow$  Option (A) is correct.

Option (B) is incorrect (obvious)



For continuous function

Each interval will have 4 choices.

- ⇒ Number of continuous functions
- $= 4 \times 4 \times 4 = 64$
- $\Rightarrow$  Option (C) is correct.
- (D) Every continuous function is piecewise constant functions
- ⇒ Differentiable.

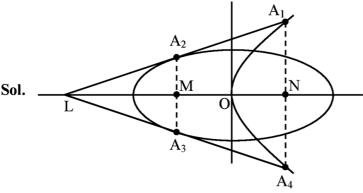
Option (D) is correct.

Let T<sub>1</sub> and T<sub>2</sub> be two distinct common tangents to the ellipse E:  $\frac{x^2}{6} + \frac{y^2}{3} = 1$  and the parabola 2.

 $P:y^2=12x$ . Suppose that the tangent  $T_1$  touches P and E at the point  $A_1$  and  $A_2$ , respectively and the tangent T<sub>2</sub> touches P and E at the points A<sub>4</sub> and A<sub>3</sub>, respectively. Then which of the following statements is(are) true?

- (A) The area of the quadrilateral  $A_1A_2A_3A_4$  is 35 square units
- (B) The area of the quadrilateral  $A_1A_2A_3A_4$  is 36 square units
- (C) The tangents  $T_1$  and  $T_2$  meet the x-axis at the point (-3, 0)
- (D) The tangents  $T_1$  and  $T_2$  meet the x-axis at the point (-6, 0)

Ans. (AC)



$$y = mx + \frac{3}{m}$$

$$C^{2} = a^{2}m^{2} + b^{2}$$

$$\frac{9}{m^{2}} = 6m^{2} + 3 \qquad \Rightarrow m^{2} = 1$$

$$T_{1} \& T_{2}$$

$$y = x + 3, y = -x - 3$$

$$Cuts x-axis at (-3, 0)$$

$$A_{1}(3, 6) \qquad A_{4}(3, -6)$$

$$A_{2}(-2, 1) \qquad A_{3}(-2, -1)$$

$$A_{1}A_{4} = 12, \quad A_{2}A_{3} = 2, \quad MN = 5$$

$$Area = \frac{1}{2}(12 + 2) \times 5 = 35 \text{ sq. unit}$$

$$Ans. (A, C)$$

- 3. Let  $f:[0, 1] \to [0, 1]$  be the function defined by  $f(x) = \frac{x^3}{3} x^2 + \frac{5}{9}x + \frac{17}{36}$ . Consider the square region  $S = [0, 1] \times [0, 1]$ . Let  $G = \{(x, y) \in S : y > f(x)\}$  be called the green region and  $R = \{(x, y) \in S : y < f(x)\}$  be called the red region. Let  $L_h = \{(x, h) \in S : x \in [0, 1]\}$  be the horizontal line drawn at a height  $h \in [0, 1]$ . Then which of the following statements is(are) ture?
  - (A) There exists an  $h \in \left\lfloor \frac{1}{4}, \frac{2}{3} \right\rfloor$  such that the area of the green region above the line  $L_h$  equals the area of the green region below the line  $L_h$
  - (B) There exists an  $h \in \left[\frac{1}{4}, \frac{2}{3}\right]$  such that the area of the red region above the line  $L_h$  equals the area of the red region below the line  $L_h$
  - (C) There exists an  $h \in \left\lfloor \frac{1}{4}, \frac{2}{3} \right\rfloor$  such that the area of the green region above the line  $L_h$  equals the area of the red region below the line  $L_h$
  - (D) There exists an  $h \in \left[\frac{1}{4}, \frac{2}{3}\right]$  such that the area of the red region above the line  $L_h$  equals the area of the green region below the line  $L_h$

Ans. (BCD)

**Sol.** 
$$f(x) = \frac{x^3}{3} - x^2 + \frac{5x}{9} + \frac{17}{36}$$

$$f'(x) = x^2 - 2x + \frac{5}{9}$$

$$f'(x) = 0$$
 at  $x = \frac{1}{3}$  in [0, 1]

 $A_R$  = Area of Red region  $A_G$  = Area of Green region

$$A_{R} = \int_{0}^{1} f(x) dx = \frac{1}{2}$$

Total area = 1

$$\Rightarrow$$
 A<sub>G</sub> =  $\frac{1}{2}$ 

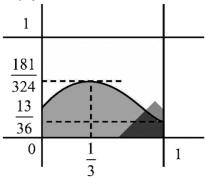
$$\int_0^1 f(x) dx = \frac{1}{2}$$

$$A_G = A_R$$

$$f(0) = \frac{17}{36}$$

$$f(1) = \frac{13}{36} \approx 0.36$$

$$f\left(\frac{1}{3}\right) = \frac{181}{324} \approx 0.558$$



- (A) Correct when  $h = \frac{3}{4}$  but  $h \in \left[\frac{1}{4}, \frac{2}{3}\right]$
- $\Rightarrow$  (A) is incorrect
- (B) Correct when  $h = \frac{1}{4}$
- $\Rightarrow$  (B) is correct
- (C) When  $h = \frac{181}{324}$ ,  $A_R = \frac{1}{2}$ ,  $A_G < \frac{1}{2}$  $h = \frac{13}{36}, A_R < \frac{1}{2}, A_G = \frac{1}{2}$
- $\Rightarrow$  A<sub>R</sub> = A<sub>G</sub> for some h  $\in$   $\left(\frac{13}{36}, \frac{181}{324}\right)$
- $\Rightarrow$  (C) is correct
- (D) Option (D) is remaining coloured part of option (C), hence option (D) is also correct.

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#### **SECTION-2: (Maximum Marks: 12)**

- This section contains **FOUR (04)** questions.
- Each question has **FOUR** options (A), (B), (C) and (D). **ONLY ONE** of these four options is the correct answer.
- For each question, choose the option corresponding to the correct answer.
- Answer to each question will be evaluated <u>according to the following marking scheme</u>:

Full Marks : +3 If **ONLY** the correct option is chosen;

Zero Marks : 0 If none of the options is chosen (i.e. the question is unanswered);

Negative Marks : -1 In all other cases.

4. Let  $f:(0, 1) \to \mathbb{R}$  be the functions defined as  $f(x) = \sqrt{n}$  if  $x \in \left[\frac{1}{n+1}, \frac{1}{n}\right]$  where  $n \in \mathbb{N}$ . Let

 $g:(0,\,1)\to\mathbb{R} \text{ be a function such that } \int\limits_{x^2}^x \sqrt{\frac{1-t}{t}}\,dt < g(x) < 2\sqrt{x} \text{ for all } x\in(0,\,1). \text{ Then } \lim_{x\to 0} f(x)g(x)$ 

(A) does NOT exist

(B) is equal to 1

(C) is equal to 2

(D) is equal to 3

Ans. (C)

Sol. 
$$\int_{x^{2}}^{x} \sqrt{\frac{1-t}{t}} dt . \sqrt{n} \le f(x)g(x) \le 2\sqrt{x} \sqrt{n}$$

$$\therefore \int_{x^{2}}^{x} \sqrt{\frac{1-t}{t}} dt = \sin^{-1} \sqrt{x} + \sqrt{x} \sqrt{1-x} - \sin^{-1} x - x\sqrt{1-x^{2}}$$

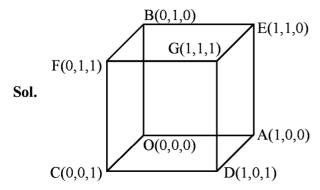
$$\Rightarrow \lim_{x \to 0} \left( \frac{\sin^{-1} \sqrt{x} + \sqrt{x} \sqrt{1-x} - \sin^{-1} x - x\sqrt{1-x^{2}}}{\sqrt{x}} \le f(x)g(x) \le \frac{2\sqrt{x}}{\sqrt{x}} \right)$$

$$\Rightarrow 2 \le \lim_{x \to 0} f(x)g(x) \le 2$$

$$\Rightarrow \lim_{x\to 0} f(x)g(x) = 2$$

- 5. Let Q be the cube with the set of vertices  $\{(x_1, x_2, x_3) \in \mathbb{R}^3 : x_1, x_2, x_3\{0,1\}\}$ . Let F be the set of all twelve lines containing the diagonals of the six faces of the cube Q. Let S be the set of all four lines containing the main diagonals of the cube Q; for instance, the line passing through the vertices (0, 0, 0) and (1, 1, 1) is in S. For lines  $\ell_1$  and  $\ell_2$ , let  $d(\ell_1, \ell_2)$  denote the shortest distance between them. Then the maximum value of  $d(\ell_1, \ell_2)$ , as  $\ell_1$  varies over F and  $\ell_2$  varies over S, is
  - $(A) \frac{1}{\sqrt{6}}$
- (B)  $\frac{1}{\sqrt{8}}$
- (C)  $\frac{1}{\sqrt{3}}$
- (D)  $\frac{1}{\sqrt{12}}$

Ans. (A)



$$DR'S \text{ of } OG = 1, 1, 1$$

DR'S of AF = 
$$-1$$
, 1, 1

DR'S of 
$$CE = 1, 1, -1$$

$$DR'S \text{ of } BD = 1, -1, 1$$

Equation of OG 
$$\Rightarrow \frac{x}{1} = \frac{y}{1} = \frac{z}{1}$$

Equation of AB 
$$\Rightarrow \frac{x-1}{1} = \frac{y}{-1} = \frac{z}{0}$$

Normal to both the line's

$$= \begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ 1 & 1 & 1 \\ 1 & -1 & 0 \end{vmatrix} = \hat{i} + \hat{j} - 2\hat{k}$$

$$\overrightarrow{OA} = \hat{i}$$

S.D. = 
$$\frac{\left|\hat{i}.(\hat{i}+\hat{j}-2\hat{k})\right|}{\left|\hat{i}+\hat{j}-2\hat{k}\right|} = \frac{1}{\sqrt{6}}$$

Ans. (A)

6. Let 
$$X = \left\{ (x,y) \in \mathbb{Z} \times \mathbb{Z} : \frac{x^2}{8} + \frac{y^2}{20} < 1 \text{ and } y^2 < 5x \right\}$$
. Three distinct points P, Q and R are randomly chosen from X. Then the probability that P, Q and R form a triangle whose area is a positive integer, is

(A) 
$$\frac{71}{220}$$

(B) 
$$\frac{73}{220}$$

(C) 
$$\frac{79}{220}$$

(D) 
$$\frac{83}{220}$$

Ans. (B)



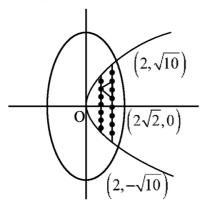
**Sol.** 
$$\frac{x^2}{8} + \frac{y^2}{20} < 1 \& y^2 < 5x$$

Solving corresponding equations

$$\frac{x^2}{8} + \frac{y^2}{20} = 1 \& y^2 = 5x$$

$$\Rightarrow \begin{cases} x = 2 \\ y = \pm \sqrt{10} \end{cases}$$

 $X = \{(1,1), (1,0), (1,-1), (1,2), (1,-2), (2,3), (2,2), (2,1), (2,0), (2,-1), (2,-2), (2,-3)\}$ 



Let S be the sample space & E be the event  $n(S) = {}^{12}C_3$ 

#### For E

Selecting 3 points in which 2 points are either or x = 1 & x = 2 but distance b/w then is even

Triangles with base 2:

$$=3\times7+5\times5=46$$

Triangles with base 4:

$$= 1 \times 7 + 3 \times 5 = 22$$

Triangles with base 6:

$$= 1 \times 5 = 5$$

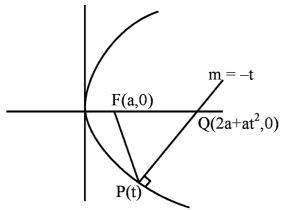
$$P(E) = \frac{46 + 22 + 5}{{}^{12}C_3} = \frac{73}{220}$$

Ans. (B)

- 7. Let P be a point on the parabola  $y^2 = 4ax$ , where a > 0. The normal to the parabola at P meets the x-axis at a point Q. The area of the triangle PFQ, where F is the focus of the parabola, is 120. If the slope m of the normal and a are both positive integers, then the pair (a,m) is
  - (A)(2,3)
- (B)(1,3)
- (C)(2,4)
- (D)(3,4)

Ans. (A)

Sol. Let point P (at<sup>2</sup>, 2at) normal at P is  $y = -tx + 2at + at^3$   $y = 0, x = 2a + at^2$  $Q(2a + at^2, 0)$ 



Area of 
$$\triangle PFQ = \left| \frac{1}{2} (a + at^2)(2at) \right| = 120$$

$$: m = -t$$

$$a^2 [1 + m^2] m = 120$$

$$(a, m) = (2, 3)$$
 will satisfy





#### **SECTION-3: (Maximum Marks: 24)**

- This section contains SIX (06) questions.
- The answer to each question is a **NON-NEGATIVE INTEGER**.
- For each question, enter the correct integer corresponding to the answer using the mouse and the on-screen virtual numeric keypad in the place designated to enter the answer.
- Answer to each question will be evaluated according to the following marking scheme:

Full Marks : +4 ONLY If the correct integer is entered;

Zero Marks : 0 In all other cases.

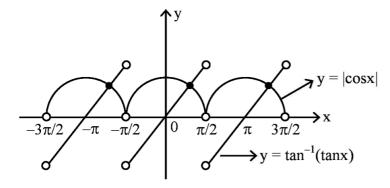
**8.** Let  $\tan^{-1}(x) \in \left(-\frac{\pi}{2}, \frac{\pi}{2}\right)$ , for  $x \in \mathbb{R}$ . Then the number of real solutions of the equation

$$\sqrt{1+\cos(2x)} = \sqrt{2}\tan^{-1}(\tan x) \text{ in the set } \left(-\frac{3\pi}{2}, -\frac{\pi}{2}\right) \cup \left(-\frac{\pi}{2}, \frac{\pi}{2}\right) \cup \left(\frac{\pi}{2}, \frac{3\pi}{2}\right) \text{ is equal to }$$

Ans. (3)

**Sol.** 
$$\sqrt{2} |\cos x| = \sqrt{2} \cdot \tan^{-1} (\tan x)$$

$$|\cos x| = \tan^{-1} \tan x$$



No. of solutions = 3

**9.** Let  $n \ge 2$  be a natural number and  $f: [0,1] \to \mathbb{R}$  be the function defined by

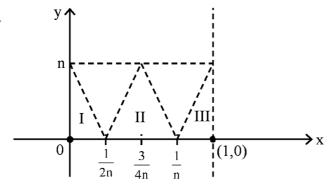
$$f(x) = \begin{cases} n(1-2nx) & \text{if } 0 \le x \le \frac{1}{2n} \\ 2n(2nx-1) & \text{if } \frac{1}{2n} \le x \le \frac{3}{4n} \\ 4n(1-nx) & \text{if } \frac{3}{4n} \le x \le \frac{1}{n} \\ \frac{n}{n-1}(nx-1) & \text{if } \frac{1}{n} \le x \le 1 \end{cases}$$

If n is such that the area of the region bounded by the curves x = 0, x = 1, y = 0 and y = f(x) is 4, then the maximum value of the function f is

Ans. (8)



Sol.



Area = Area of (I + II + III) = 4

$$= \frac{1}{2} \times \frac{1}{2n} \times n + \frac{1}{2} \times \frac{1}{2n} \times n + \frac{1}{2} \left( 1 - \frac{1}{n} \right) \times n$$
$$= \frac{1}{4} + \frac{1}{4} + \frac{n-1}{2} = 4$$

$$n = 8$$

 $\therefore$  maximum value of f(x) = 8

10. Let 75...57 denote the (r + 2) digit number where the first and the last digits are 7 and the remaining r digits are 5. Consider the sum S = 77 + 757 + 7557 + ... + 75...57. If  $S = \frac{75...57 + m}{n}$ , where m and n are natural numbers less than 3000, then the value of m + n is

Ans. (1219)

**Sol.** 
$$S = 77 + 757 + 7557 + ... + 75.....57$$

$$10S = 770 + 7570 + ... + 75 ... 570 + 755 .....570$$

$$9S = -77 + \underbrace{13 + 13 + \dots + 13}_{98 \text{ times}} + 75 \dots 570$$
$$= -77 + 13 \times 98 + 75 \dots 57 + 13$$

$$S = \frac{75.....57 + 1210}{99}$$

$$m = 1210$$

$$n = 9$$

$$m + n = 1219$$

11. Let  $A = \left\{ \frac{1967 + 1686i \sin \theta}{7 - 3i \cos \theta} : \theta \in \mathbb{R} \right\}$ . If A contains exactly one positive integer n, then the value of

n is

Ans. (281)

Sol. 
$$A = \frac{1967 + 1686i\sin\theta}{7 - 3i\cos\theta}$$
$$= \frac{281(7 + 6i\sin\theta)}{7 - 3i\cos\theta} \times \frac{7 + 3i\cos\theta}{7 + 3i\cos\theta}$$
$$= \frac{281(49 - 18\sin\theta\cos\theta + i(21\cos\theta + 42\sin\theta))}{49 + 9\cos^2\theta}$$

for positive integer

$$Im(A) = 0$$

$$21\cos\theta + 42\sin\theta = 0$$

$$\tan \theta = \frac{-1}{2}$$
;  $\sin 2\theta = \frac{-4}{5}$ ,  $\cos^2 \theta = \frac{4}{5}$ 

$$Re(A) = \frac{281(49 - 9\sin 2\theta)}{49 + 9\cos^2 \theta}$$

$$= \frac{281\left(49 - 9 \times \frac{-4}{5}\right)}{49 + 9 \times \frac{4}{5}} = 281 \text{ (+ve integer)}$$

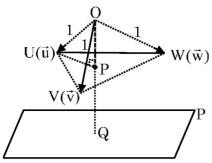
12. Let P be the plane  $\sqrt{3}x + 2y + 3z = 16$  and let

$$S = \left\{ \alpha \hat{i} + \beta \hat{j} + \gamma \hat{k} : \alpha^2 + \beta^2 + \gamma^2 = 1 \text{ and the distance of } (\alpha, \beta, \gamma) \text{ from the plane P is } \frac{7}{2} \right\}.$$

Let  $\vec{u}$ ,  $\vec{v}$  and  $\vec{w}$  be three distinct vectors in S such that  $|\vec{u} - \vec{v}| = |\vec{v} - \vec{w}| = |\vec{w} - \vec{u}|$ . Let V be the volume of the parallelepiped determined by vectors  $\vec{u}$ ,  $\vec{v}$  and  $\vec{w}$ . Then the value of  $\frac{80}{\sqrt{3}}$  V is

Ans. (45)

Sol.



Given  $|\vec{u} - \vec{v}| = |\vec{v} - \vec{w}| = |\vec{w} - \vec{u}|$ 

 $\Rightarrow \Delta UVW$  is an equilateral  $\Delta$ 





Now distances of U, V, W from  $P = \frac{7}{2}$ 

$$\Rightarrow$$
 PQ =  $\frac{7}{2}$ 

Also, Distance of plane P from origin

$$\Rightarrow$$
 OQ = 4

$$\therefore OP = OQ - PQ \Rightarrow OP = \frac{1}{2}$$

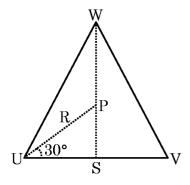
Hence, 
$$PU = \sqrt{OU^2 - OP^2} \implies PU = \frac{\sqrt{3}}{2} = R$$

Also, for ΔUVW, P is circumcenter

$$\therefore$$
 for  $\triangle UVW$ : US = Rcos30°

$$\Rightarrow$$
 UV = 2Rcos30°

$$\Rightarrow$$
 UV =  $\frac{3}{2}$ 



$$\therefore Ar(\Delta UVW) = \frac{\sqrt{3}}{4} \left(\frac{3}{2}\right)^2 = \frac{9\sqrt{3}}{16}$$

 $\therefore$  Volume of tetrahedron with coterminous edges  $\vec{u},\,\vec{v},\vec{w}$ 

$$=\frac{1}{3}(Ar.\Delta UVW)\times OP = \frac{1}{3}\times \frac{9\sqrt{3}}{16}\times \frac{1}{2} = \frac{3\sqrt{3}}{32}$$

... parallelopiped with coterminous edges

$$\vec{u}$$
,  $\vec{v}$ ,  $\vec{w} = 6 \times \frac{3\sqrt{3}}{32} = \frac{9\sqrt{3}}{16} = V$ 

$$\therefore \frac{80}{\sqrt{3}} V = 45$$

### IIT-JEE ADVANCED 2023 PAPER-1

13. Let a and b be two nonzero real numbers. If the coefficient of  $x^5$  in the expansion of  $\left(ax^2 + \frac{70}{27bx}\right)^4$  is equal to the coefficient of  $\left(ax - \frac{1}{bx^2}\right)^7$ , then the value of 2b is

Ans. (3)

Sol. 
$$T_{r+1} = {}^{4}C_{r} (a.x^{2})^{4-r} . \left(\frac{70}{27bx}\right)^{r}$$
$$= {}^{4}C_{r} . a^{4-r} . \frac{70^{r}}{(27b)^{r}} . x^{8-3r}$$

here 
$$8 - 3r = 5$$

$$8-5=3r \Rightarrow r=1$$

$$\therefore coeff. = 4.a^3. \frac{70}{27b}$$

$$T_{r+1} = {}^{7}C_{r}(ax)^{7-r} \left(\frac{-1}{bx^{2}}\right)^{r}$$

$$= {}^{7}C_{r}.a^{7-r} \left(\frac{-1}{b}\right)^{r}.x^{7-3r}$$

$$7 - 3r = -5 \Rightarrow 12 = 3r \Rightarrow r = 4$$

coeff.: 
$${}^{7}C_{4}.a^{3}.\left(\frac{-1}{b}\right)^{4} = \frac{35a^{3}}{b^{4}}$$

$$now \ \frac{35a^3}{b^4} = \frac{280a^3}{27b}$$

$$b^3 = \frac{35 \times 27}{280} = b = \frac{3}{2} \implies 2b = 3$$

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#### **SECTION-4: (Maximum Marks: 12)**

- This section contains FOUR (04) Matching List Sets.
- Each set has **ONE** Multiple Choice Question.
- Each set has TWO lists: List-I and List-II.
- List-I has Four entries (P), (Q), (R) and (S) and List-II has Five entries (1), (2), (3), (4) and (5).
- FOUR options are given in each Multiple Choice Question based on List-I and List-II and ONLY **ONE** of these four options satisfies the condition asked in the Multiple Choice Question.
- Answer to each question will be evaluated according to the following marking scheme:

: +3 **ONLY** if the option corresponding to the correct combination is chosen; Full Marks

Zero Marks : 0 If none of the options is chosen (i.e. the question is unanswered);

Negative Marks : -1 In all other cases.

Let  $\alpha$ ,  $\beta$  and  $\gamma$  be real numbers. consider the following system of linear equations

$$x + 2y + z = 7$$

$$x + \alpha z = 11$$

$$2x - 3y + \beta z = \gamma$$

Match each entry in List - I to the correct entries in List-II

List-I List-II

(P) If 
$$\beta = \frac{1}{2}(7\alpha - 3)$$
 and  $\gamma = 28$ , then the system has

(Q) If 
$$\beta = \frac{1}{2}(7\alpha - 3)$$
 and  $\gamma \neq 28$ , then the system has

(R) If 
$$\beta \neq \frac{1}{2}$$
 (7 $\alpha$  –3) where  $\alpha = 1$  and  $\gamma \neq 28$ ,

(3) infinitely many solutions

then the system has

(S) If 
$$\beta \neq \frac{1}{2}$$
 (7 $\alpha$  – 3) where  $\alpha = 1$  and  $\gamma = 28$ ,

(4) 
$$x = 11$$
,  $y = -2$  and  $z = 0$  as a solution

then the system has

(5) 
$$x = -15$$
,  $y = 4$  and  $z = 0$  as a solution

The correct option is:

$$(A) (P) \rightarrow (3) (Q) \rightarrow (2) (R) \rightarrow (1) (S) \rightarrow (4)$$

(B) (P) 
$$\rightarrow$$
 (3) (Q)  $\rightarrow$  (2) (R)  $\rightarrow$  (5) (S)  $\rightarrow$  (4)

$$(C)(P) \rightarrow (2)(Q) \rightarrow (1)(R) \rightarrow (4)(S) \rightarrow (5)$$

(D) (P) 
$$\rightarrow$$
 (2) (Q)  $\rightarrow$  (1) (R)  $\rightarrow$  (1) (S)  $\rightarrow$  (3)

Ans. (A)

**Sol.** Given x + 2y + z = 7 .... (1)

$$x + \alpha z = 11 \qquad \dots (2)$$

$$2x - 3y + \beta z = \gamma \qquad \dots (3)$$

Now, 
$$\Delta = \begin{vmatrix} 1 & 2 & 1 \\ 1 & 0 & \alpha \\ 2 & -3 & \beta \end{vmatrix} = 7\alpha - 2\beta - 3$$

$$\therefore \text{ if } \beta = \frac{1}{2}(7\alpha - 3)$$

$$\Rightarrow \Delta = 0$$

$$\Rightarrow \boxed{\Delta = 0}$$

$$\Rightarrow \boxed{\Delta = 0}$$
Now,  $\Delta_{x} = \begin{vmatrix} 7 & 2 & 1 \\ 11 & 0 & \alpha \\ \gamma & -3 & \beta \end{vmatrix}$ 

$$= 21\alpha - 22\beta + 2\alpha\gamma - 33$$

$$=21\alpha-22\beta+2\alpha\gamma-33$$

$$\therefore$$
 if  $\gamma = 28$ 

$$\Rightarrow \Delta_{\mathbf{x}} = 0$$

$$\Delta_y = \begin{vmatrix} 1 & 7 & 1 \\ 1 & 11 & \alpha \\ 2 & \gamma & \beta \end{vmatrix}$$

$$\Delta_{v} = 4\beta + 14\alpha - \alpha\gamma + \gamma - 22$$

$$\therefore \text{ if } \gamma = 28$$

$$\Rightarrow \Delta_{y} = 0$$

Now, 
$$\Delta_z = \begin{vmatrix} 1 & 2 & 7 \\ 1 & 0 & 11 \\ 2 & -3 & \gamma \end{vmatrix} = 56 - 2\gamma$$

If 
$$\gamma = 28$$

$$\Rightarrow \Delta_z = 0$$

$$\therefore \text{ if } \gamma = 28 \text{ and } \beta = \frac{1}{2}(7\alpha - 3)$$

⇒ system has infinite solution and if  $\gamma \neq 28$ 

⇒ system has no solution

$$\Rightarrow$$
 P  $\rightarrow$  (3); Q  $\rightarrow$  (2)

Now if 
$$\beta \neq \frac{1}{2}(7\alpha - 3)$$

$$\Rightarrow \Delta \neq 0$$

and for  $\alpha = 1$  clearly

y = -2 is always be the solution

 $\therefore$  if  $\gamma \neq 28$ 

System has a unique solution

if 
$$\gamma = 28$$

$$\Rightarrow$$
 x = 11, y = -2 and z = 0 will be one of the solution

$$\therefore R \to 1; S \to 4$$

#### 15. Consider the given data with frequency distribution

 $x_i$  3 8 11 10 5 4  $f_i$  5 2 3 2 4 4

Match each entry in List-I to the correct entries in List-II.

#### List-II List-II

(P) The mean of the above data is (1) 2.5

(Q) The median of the above data is (2) 5

(R) The mean deviation about the mean of the above data is (3) 6

(S) The mean deviation about the median of the above data is (4) 2.7

(5)2.4

The correct option is:

(A) (P) 
$$\rightarrow$$
 (3) (Q)  $\rightarrow$  (2) (R)  $\rightarrow$  (4) (S)  $\rightarrow$  (5)

(B) (P) 
$$\rightarrow$$
 (3) (Q)  $\rightarrow$  (2) (R)  $\rightarrow$  (1) (S)  $\rightarrow$  (5)

$$(C)(P) \rightarrow (2)(Q) \rightarrow (3)(R) \rightarrow (4)(S) \rightarrow (1)$$

(D) (P) 
$$\rightarrow$$
 (3) (Q)  $\rightarrow$  (3) (R)  $\rightarrow$  (5) (S)  $\rightarrow$  (5)

#### Ans. (A)

**Sol.** x<sub>i</sub> 3 4 5 8 10 11

 $f_i$  5 4 4 2 2 3

(P) Mean

(Q) Median

(R) Mean deviation about mean

(S) Mean deviation about median

Xi	$f_i$	$x_i f_i$	C.F.	$ x_i - Mean $	$f_i   x_i - Mean  $	$ x_i - Median $	$f_i x_i - Median $
3	5	15	5.	3	15	2	10
4	4	16	9	2	8	1	4
5	4	20	13	1	4	0	0
8	2	16	15	2	4	3	6
10	2	20	17	4	8	5	10
11	3	33	20	5	15	6	18
	$\Sigma f_i = 20$	$\Sigma \mathbf{x_i} \mathbf{f_i} = 120$			$ \Sigma f_i x_i - Mean  = 54$		$\Sigma f_i   x_i - Median   = 48$

(P) Mean = 
$$\frac{\sum x_i f_i}{\sum f_i} = \frac{120}{20} = 6$$

(Q) Median = 
$$\left(\frac{20}{2}\right)^{th}$$
 observation =  $10^{th}$  observation = 5

(R) Mean deviation about mean = 
$$\frac{\Sigma f_i |x_i - Mean|}{\Sigma f_i} = \frac{54}{20} = 2.70$$

(S) mean deviation about median = 
$$\frac{\sum f_i |x_i - Median|}{\sum f_i} = \frac{48}{20} = 2.40$$

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16. Let  $\ell_1$  and  $\ell_2$  be the lines  $\vec{r}_1 = \lambda (\hat{i} + \hat{j} + \hat{k})$  and  $\vec{r}_2 = (\hat{j} - \hat{k}) + \mu (\hat{i} + \hat{k})$ , respectively. Let X be the set of all the planes H that contain the line  $\ell_1$ . For a plane H, let d(H) denote the smallest possible distance between the points of  $\ell_2$  and H. Let  $H_0$  be plane in X for which  $d(H_0)$  is the maximum value of d(H) as H varies over all planes in X.

Match each entry in List-I to the correct entries in List-II.

List-I

List-II

(P) The value of  $d(H_0)$  is

(1)  $\sqrt{3}$ 

(Q) The distance of the point (0,1,2) from  $H_0$  is

(2)  $\frac{1}{\sqrt{3}}$ 

(R) The distance of origin from  $H_0$  is

- (3)0
- (S) The distance of origin from the point of intersection
- (4)  $\sqrt{2}$

of planes y = z, x = 1 and  $H_0$  is

(5)  $\frac{1}{\sqrt{2}}$ 

The correct option is:

$$(A) (P) \rightarrow (2) (Q) \rightarrow (4) (R) \rightarrow (5) (S) \rightarrow (1)$$

(B) (P) 
$$\rightarrow$$
 (5) (Q)  $\rightarrow$  (4) (R)  $\rightarrow$  (3) (S)  $\rightarrow$  (1)

$$(C)(P) \rightarrow (2)(Q) \rightarrow (1)(R) \rightarrow (3)(S) \rightarrow (2)$$

(D) (P) 
$$\rightarrow$$
 (5) (Q)  $\rightarrow$  (1) (R)  $\rightarrow$  (4) (S)  $\rightarrow$  (2)

Ans. (B)

Ans. ()

**Sol.** 
$$L_1: \vec{r}_1 = \lambda(\hat{i} + \hat{j} + \hat{k})$$

$$L_2: \vec{r}_2 = \hat{j} - \hat{k} + \mu(\hat{i} + \hat{k})$$

Let system of planes are

$$ax + by + cz = 0$$

.... (1)

∵ It contain L<sub>1</sub>

$$\therefore a + b + c = 0$$

.... (2)

For largest possible distance between plane (1) and L<sub>2</sub> the line L<sub>2</sub> must be parallel to plane (1)

$$\therefore a + c = 0$$

.... (3)

$$\Rightarrow b = 0$$

 $\therefore \text{ Plane } H_0: \boxed{\mathbf{x} - \mathbf{z} = 0}$ 





Now  $d(H_0) = \bot$  distance from point (0, 1, -1) on  $L_2$  to plane.

$$\Rightarrow d(H_0) = \left| \frac{0+1}{\sqrt{2}} \right| = \frac{1}{\sqrt{2}}$$

$$\therefore P \rightarrow 5$$

for 'Q' distance = 
$$\left| \frac{2}{\sqrt{2}} \right| = \sqrt{2}$$

$$\therefore Q \rightarrow 4$$

 $\therefore$  (0, 0, 0) lies on plane

$$\therefore R \rightarrow 3$$

for 'S' 
$$x = z$$
;  $y = z$ ;  $x = 1$ 

 $\therefore$  point of intersection p(1, 1, 1).

$$\therefore OP = \sqrt{1+1+1} = \sqrt{3}$$

$$\therefore S \rightarrow 2$$

.. option [B] is correct

17. Let z be complex number satisfying  $|z|^3 + 2z^2 + 4\overline{z} - 8 = 0$ , where  $\overline{z}$  denotes the complex conjugate of z. Let the imaginary part of z be nonzero.

Match each entry in List-I to the correct entries in List-II.

#### List-I

#### List-II

(P) 
$$|z|^2$$
 is equal to

(Q) 
$$|z-\overline{z}|^2$$
 is equal to

(R) 
$$|z|^2 + |z + \overline{z}|^2$$
 is equal to

(S) 
$$|z+1|^2$$
 is equal to

The correct option is:

$$(A) (P) \rightarrow (1) (Q) \rightarrow (3) (R) \rightarrow (5) (S) \rightarrow (4)$$

(B) (P) 
$$\rightarrow$$
 (2) (Q)  $\rightarrow$  (1) (R)  $\rightarrow$  (3) (S)  $\rightarrow$  (5)

$$(C)(P) \rightarrow (2)(Q) \rightarrow (4)(R) \rightarrow (5)(S) \rightarrow (1)$$

(D) (P) 
$$\rightarrow$$
 (2) (Q)  $\rightarrow$  (3) (R)  $\rightarrow$  (5) (S)  $\rightarrow$  (4)

Ans. (B)

**PAPER-1** 

**Sol.** : 
$$|z|^3 + 2z^2 + 4\overline{z} - 8 = 0$$

....(1)

Take conjugate both sides

$$\Rightarrow |z|^3 + 2\overline{z}^2 + 4z - 8 = 0$$

By 
$$(1) - (2)$$

$$\Rightarrow 2(z^2 - \overline{z}^2) + 4(\overline{z} - z) = 0$$

$$\Rightarrow \boxed{z + \overline{z} = 2}$$

$$\Rightarrow |z + \overline{z}| = 2$$

Let 
$$z = x + iy$$

$$\therefore \mathbf{x} = 1$$

$$\therefore z = 1 + iy$$

Put in (1)

$$\Rightarrow (1+y^2)^{3/2} + 2(1-y^2 + 2iy) + 4(1-iy) - 8 = 0$$

$$\Rightarrow (1 + y^2)^{3/2} = 2(1 + y^2)$$

$$\Rightarrow \sqrt{1+y^2} = 2 = |z|$$

Also 
$$y = \pm \sqrt{3}$$

$$\therefore z = 1 \pm i\sqrt{3}$$

$$\Rightarrow$$
 z -  $\overline{z}$  =  $\pm 2i\sqrt{3}$ 

$$\Rightarrow |z - \overline{z}| = 2\sqrt{3}$$

$$\Rightarrow |z - \overline{z}|^2 = 12$$

Now 
$$z + 1 = 2 + i\sqrt{3}$$

$$|z+1|^2=4+3=7$$

$$\therefore \ P \rightarrow 2 \ ; \ Q \rightarrow 1 \ ; \ R \rightarrow 3 \ ; \ S \rightarrow 5$$

.. Option [B] is correct.